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(54) **Metallic water borne base coat composition.**

(57) There is disclosed a metallic and nonmetallic water borne base coat compositions exhibiting excellent appearance. A method for long term stability and the introduction of aluminum flake pigments into a water borne coating is also disclosed. The method is based on storing the aluminum flake in a solvent environment with introduction of the aluminum flake in to the aqueous environment just before applying the base coat coating. A water reducible acrylic resin with nonionic and acid functionality has been disclosed that allows for direct introduction of aluminum flake in to an aqueous environment. The coating vehicle is based on acrylic latex vehicle which exhibits excellent drying and film properties. The coating compositions disclosed are in compliance with current and proposed volatile organic compound regulations.

EP 0 471 972 A2

## FIELD OF INVENTION

The present invention is concerned with a water borne coating composition designed for the coating of transportation vehicles. The invention is related to coating compositions primarily for a base coat of a multi-coat system which includes primers and transparent clear coats, particularly with base coats containing metallic pigments, based on acrylic latex polymers.

## BACKGROUND AND PRIOR ART

Related automotive paint compositions of this type are disclosed in U.S. patent applications titled METALLIC WATER BORNE BASE COAT COMPOSITION BASED ON ACRYLIC LATEX RESINS USING A WATER REDUCIBLE RESIN FOR ALUMINUM STORAGE AND A PREGELLED HECTORITE CLAY FOR RHEOLOGY CONTROL Ser. No. filed, METALLIC WATER BORNE BASE COAT COMPOSITION BASED ON ACRYLIC LATEX RESINS USING A WATER REDUCIBLE RESIN FOR ALUMINUM STORAGE AND A TREATED HECTORITE CLAY FOR RHEOLOGY CONTROL Ser. No. filed, METALLIC WATER BORNE BASE COAT COMPOSITION BASED ON ACRYLIC LATEX RESINS USING AN ACRYLIC RESIN HAVING ACID AND NONIONIC FUNCTIONALITY FOR ALUMINUM STORAGE AND A PREGELLED HECTORITE CLAY FOR RHEOLOGY CONTROL Ser. No. filed, METALLIC WATER BORNE BASE COAT COMPOSITION BASED ON ACRYLIC LATEX RESINS USING AN ACRYLIC RESIN HAVING ACID AND NONIONIC FUNCTIONALITY FOR ALUMINUM STORAGE AND A TREATED HECTORITE CLAY FOR RHEOLOGY CONTROL Ser. No. filed, METALLIC WATER BORNE BASE COAT COMPOSITION BASED ON ACRYLIC LATEX RESINS AND ACRYLIC RESINS DERIVED FROM ACRYLIC MONOMERS HAVING NONIONIC, URETHANE AND ACID FUNCTIONALITY FOR ALUMINUM STORAGE AND A PREGELLED HECTORITE CLAY FOR RHEOLOGY CONTROL Ser. No. filed, METALLIC WATER BORNE BASE COAT COMPOSITION BASED ON ACRYLIC LATEX RESINS USING ACRYLIC RESINS DERIVED FROM VINYL MONOMERS HAVING NONIONIC AND URETHANE FUNCTIONALITY FOR ALUMINUM STORAGE AND A PREGELLED HECTORITE CLAY FOR RHEOLOGY CONTROL Ser. No. filed, METALLIC WATER BORNE BASE COAT COMPOSITION BASED ON ACRYLIC LATEX RESINS USING ACRYLIC RESINS DERIVED FROM VINYL MONOMERS HAVING NONIONIC AND URETHANE FUNCTIONALITY FOR ALUMINUM STORAGE AND A TREATED HECTORITE CLAY FOR RHEOLOGY CONTROL Ser. No. filed, METALLIC WATER BORNE BASE COAT COMPOSITION BASED ON ACRYLIC LATEX RESINS USING ACRYLIC RESINS DERIVED FROM GRAFTING AN ISOCYANATE ADDUCT HAVING URETHANE AND NONIONIC FUNCTIONALITY ON TO AN ACRYLIC RESIN HAVING ACTIVE HYDROGEN FUNCTIONALITY FOR ALUMINUM STORAGE AND A PREGELLED HECTORITE CLAY FOR RHEOLOGY CONTROL Ser. No. filed, METALLIC WATER BORNE BASE COAT COMPOSITION BASED ON ACRYLIC LATEX RESINS USING ACRYLIC RESINS DERIVED FROM GRAFTING AN ISOCYANATE ADDUCT HAVING URETHANE AND NONIONIC FUNCTIONALITY ON TO AN ACRYLIC RESIN HAVING ACTIVE HYDROGEN FUNCTIONALITY FOR ALUMINUM STORAGE AND A TREATED HECTORITE CLAY FOR RHEOLOGY CONTROL Ser. No. filed, and ACRYLIC LATEX METALLIC WATER BORNE BASE COAT OF IMPROVED APPEARANCE Ser. No. 07/355,959 filed May 23, 1989. All of these U.S. patents and applications are incorporated herein by reference.

It is customary in the painting of an automobile that

a series of coatings be applied to the substrate. The first coat being the primer followed by the base coat and finally the clear coat. The base coat provides the good decorative quality to the final finish via organic and inorganic pigments. In many automobile finishes, a metallic finish is desired. To obtain this metallic effect, metallic pigments are present in the base coat, typically aluminum flakes. The aluminum flakes in the base coat contribute to the glossy lustrous appearance of the final finish.

Another feature that aluminum flakes contribute to the appearance of the final finish is the change of brightness as the viewing angle is changed. Typically, the final finish is bright when viewed at a 90 degree angle, however, when the viewing angle is changed to 180 degrees, the coating has a dark appearance. This phenomenon is known as "flop" in the automotive field and is highly desired. The ultimate in "flop" is achieved when the aluminum flakes are orientated parallel to the substrate. Random orientation of the aluminum flake results in a final finish exhibiting bright and dark areas when viewed at 90 degrees. This appearance is known as mottling and detracts from the final appearance of the final finish. In general, the metal fixation in base coats is achieved by a rheology modifiers such as inorganic and organic thickeners. The particular rheology control agent used in this invention results in a non-mottled, high head-on-

brightness, outstanding flop, and high quality finish, even in the case of a silver metallic base coat.

In the current market place, automobile coatings, especially base coats, contain a high level of organic solvent. With increasing concern about the volatile organic emissions into the atmosphere, an intensive effort in research and development of coatings containing mainly water as the solvent with a small level of organic solvent is under way. An example of such an effort is U.S. patent No. 4,730,020 which discloses a water-dilutable coating composition comprising specifically selected acrylic copolymers, solvent blends, coloring and/or optical effect pigments and polymer dispersions. To obtain the desired optical effect of the metallic flakes, the correct combination of acrylic copolymer and solvent blend must be achieved. An aqueous thermosetting acrylic resin described by U.S. Patent No. 3,862,071 controls the metallic pigment orientation by the addition of a water insoluble copolymer. Microgel technology as described by GB-PS No. 2,073,609 also results in the proper metal orientation. Also disclosed in DE No. 3,210,051 is an attempt to control metallic pigment orientation using polyurethane dispersions. Cellulosic esters have also been used to control metal fixation as disclosed in DE No 3,216,549. The rheology modifiers or rheology control agents for water borne coatings have poor shelf stability, poor weathering characteristics and are cumbersome to use. The particular rheology control agent used in this invention results in a non-mottled, high head-on-brightness, outstanding flop, and high quality finish, even in the case of a silver metallic base coat and does not suffer from the problems cited earlier.

The use of acrylic latex resin in a base coat formulation has been disclosed in European Application 0 287 144 A1, however, this application does not disclose any rheology control agents, pigment dispersion procedures, method for aluminum storage and requires that the acrylic latex resin be prepared in 2 or more steps by emulsion polymerization.

Typically, coating compositions used in the automotive market, especially in the automotive after market, are produced by mixing various bases to give the desired color. These coating compositions are then applied in about 1-5 days after preparation. A major problem is the introduction of aluminum flakes which react with water to generate hydrogen gas. Therefore, the aluminum flake must be segregated from the rest of the water borne components to minimize this hazard.

This invention also describes a method for the storage of aluminum flake in an organic medium with introduction of the flakes directly into the aqueous environment without neutralization just prior to application of the coating system.

### SUMMARY OF INVENTION

This invention relates to water borne compositions for use in metallic and nonmetallic base coats in a multicoat system for the automotive market. Typically, a water borne basecoat coating composition for the refinish market are in a kit form. The present invention is concerned with an acrylic latex coating composition for use in an automotive paint base coat composition. The invention also relates to base coats containing pigments, metallic pigments, organic solvents, and conventional paint additives. The outstanding metal control exhibited by this coating composition is attributed to the rheology control agent, the film shrinkage of the acrylic latex vehicle while drying and a wax. The use of acrylic latex results in a very fast dry time. This coating composition provides a base coat that satisfies current and proposed volatile organic compound regulations. Even with conventional non-metallic pigments, the coating exhibits excellent appearance.

This invention describes a method for the storage of aluminum flake in an organic medium with introduction of the flakes into the aqueous environment just prior to application of the coating system. This method minimizes the potential hazard of the aluminum flake reacting with water to form hydrogen gas and dulling of the aluminum flake during storage. This method can also be used for the storage of other water sensitive pigments such as copper and brass. Other pigments such as plastic films, mica and coated metallic foils can be stored using this method.

The automotive metallic base coat paint comprises three components; A) aluminum base, B) pigmented or unpigmented base, and C) reducer base. The automotive nonmetallic base coat paint comprises two components; A) pigmented base and B) reducer base. Component A, the aluminum base, is composed of metallic flakes slurred into a solvent borne acrylic resin and water miscible solvents. The acrylic resin used in this component has nonionic and acid functionality allowing for direct introduction of the component into aqueous acrylic latex resin system. The acrylic resin used in this component preferably has nonionic, urethane and acid functionality. The aluminum dispersion resin is an acrylic polymer in water-miscible solvent(s) and can be formed by the free-radical initiated polymerization of ethylenically unsaturated monomers. A polyether containing ethylenically unsaturated monomer and/or a novel polyether-urethane containing ethylenically unsaturated monomer is included in the monomer mixture, to give a final poly-

(ethylene oxide) level of 1-50% by weight, preferably 5-20% by weight. The number average molecular weight of the polymer is between 1,000 and 50,000 atomic mass units, its Glass Transition Temperature ( $T_g$ ) is between  $-20^\circ$  and  $+50^\circ$  C (calculated by the Fox method), a hydroxyl number of 0-50, and its acid number is from 5 to 50 milligrams of KOH per gram of resin solids. The viscosity of the final polymer is 3,000-10,000 cps with a solids content of 60-70% by weight at  $23^\circ$  C.

The aluminum dispersion resin also can be prepared by the reaction of an acrylic resin having active hydrogen functionality with an isocyanate adduct having urethane and nonionic functionality. The final resin has a final poly(ethylene oxide) level of 1 - 50 % by weight, preferably 5 - 20 % by weight. The number average molecular weight of the polymer is between 1,000 and 50,000 atomic mass units, its Glass Transition Temperature ( $T_g$ ) is between  $-20^\circ$  C and  $+50^\circ$  C (calculated by the Fox method), a hydroxyl number of 0 - 50, and its acid number is from 5 to 50 milligrams of KOH per gram of resin solids. The viscosity of the final polymer is 3,000 - 10,000 cps with a solids content of 60 - 70 % by weight at  $23^\circ$  C.

Commonly used metallic pigments, used singularly or as a mixture are copper, copper alloys, aluminum, steel, mica, plastic films, and coated metallic foils, preferably aluminum at a level of 4-30% by weight, preferably 20-30% by weight. The aluminum base contains 5-35%, preferably 20-30% by weight of the solvent borne water reducible acrylic resin. The organic solvent for this base must be water miscible. Examples of solvents are methanol, ethanol, propanol, butanol, N-methylpyrrolidone, glycols, glycol ethers, glycol acetates, diethylene glycol ethers, diethylene glycol acetates, propylene glycol ethers, propylene glycol acetates, dipropylene glycol ethers, dipropylene glycol acetates, specially preferred are ethylene glycol propyl ether, ethylene glycol butyl ether, ethylene glycol hexyl ether, propylene methyl ether, propylene ethyl ether, propylene propyl ether, propylene butyl ether, propylene hexyl ether, dipropylene methyl ether, dipropylene ethyl ether, dipropylene propyl ether, and dipropylene butyl ether.

Component A, the aluminum base, has a total solids level of 25-55 by weight, total volatile organic solvent level of 35-50% by weight, total metallic pigment level of 4-30% by weight and total binder level of 5-35% by weight.

By binder we mean the nonvolatile portion of the resinous vehicle of a coating, in particular reference is given to Paint & Coatings Dictionary, Published by The Federation of Societies for Coatings Technology, 1978. By solids we mean the nonvolatile matter in a coating composition left behind after drying, in particular reference is given to Paint & Coatings Dictionary, Published by the Federation of Societies for Coatings Technology, 1978.

Component B, pigmented base, contains acrylic latex grinding polymer and acrylic latex film forming vehicle at a level of 5-50%, preferably 15-50%. This level depends on the pigment in the component. The level of pigment in component B is 0.5-35% by weight, this level depends on the pigment characteristics also. Typically, a pigment to binder weight ratio of about 10/100 to 300/100 is acceptable. The pigments are typically ground using conventional dispersion equipment such as sand mills, pearl mills, ball mills, horizontal mills, and vertical mills. Optional wetting agents, surfactants, and dispersing aids can be employed. This component may also contain a wax composed of ethylene and/or ethylenically unsaturated monomers alkyl (meth)acrylates such as methyl (meth)acrylate, ethyl (meth)acrylate, propyl (meth)acrylate, 2-propyl (meth)acrylate, butyl (meth)acrylate, 2-butyl(meth)acrylate, 2-methylpropyl (meth)acrylate, t-butyl (meth)acrylate, 2-ethylhexyl (meth)acrylate, cyclohexyl (meth)acrylate, lauryl (meth)acrylate, octyl (meth)acrylate, decyl (meth)acrylate, stearyl (meth)acrylate, dicyclopentadienyl (meth)acrylate, and isobornyl (meth)acrylate; aryl monomers such as styrene,  $\alpha$ -methylstyrene, vinyl toluene, t-butyl styrene, benzyl (meth)acrylate; and 2-benzylethyl (meth)acrylate; hydroxy monomers such as hydroxyethyl (meth)acrylate, hydroxypropyl (meth)acrylate, hydroxybutyl (meth)acrylate, sec-hydroxybutyl (meth)acrylate, and hydroxyoctyl (meth)acrylate; acid functional monomers such as acrylic acid, methacrylic acid, maleic acid, and itaconic acid. Typically, the particle size of the wax particle in the final dispersion is 0.5 microns to 50 microns, preferably 1.0 to 20.0 microns. The wax is dispersed by conventional techniques or emulsified in to the resin system to the desired particle size. The proper amount of the wax dispersion is then added to the pigmented base. The pH of the pigmented base can be adjusted using ammonia or any water soluble primary amines, secondary amines, tertiary amines, polyamines and hydroxyamines, such as ethanolamine, diethanolamine, triethanolamine, n-methylethanol amine, N,N diethylethanolamine, N-aminoethanolamine, N-methyldiethanolamine, monoisopropanolamine, diisopropanolamine, triisopropanolamine, hydroxyamine, butanolamine, hexanolamine, methyl diethanolamine, N,N-diethylaminoethylamine, ethylene diamine, diethylene triamine, diethylene tetraamine, hexamethylene tetraamine, triethylamine and the like. The preferred base is ammonia. This component may also contain a water dispersible polyurethane resin at a level of 0-25% by weight.

Component B, the pigmented or unpigmented base, has a total solids level of 5-60 % by weight, a total pigment level of 0.0-45 % by weight, a total organic solvent level of 0-25 % by weight, a total binder of 20-

60 % by weight, a total wax level of 0-20 % by weight, a total water level of 40-95 % by weight, a pH range of 7-9 and a total water dispersible polyurethane resin at a level of 0-25 % by weight.

Component C, reducer base, contains an aqueous solution of; 1) synthetic sodium lithium magnesium silicate hectorite clay at a level of 0.1-10 % by weight, preferable 0.5-3.0 % by weight, 2) a polypropylene glycol at a level of 0.1-10 % by weight and/or an anionic surfactant at a level of 0.1-2.0 % by weight, preferable 0.1-1.0 % by weight, 3) a polyurethane resin at a level of 0.0 to 25.0 % by weight, preferable 0.0 to 10.0 %, 4) water miscible solvent at a level of 0-25 % by weight, 5) a total solids level of 0.1-20 % by weight and 6) a total water level of 50.0-99.9 % by weight. Component B, pigmented base, contains acrylic latex grinding polymer and acrylic latex film forming vehicle at a level of 5-50 %, preferably 15-50 %. This level depends on the pigment in the component. This level of pigment in component B is 0.5-35 % by weight, this level depends on the pigment characteristics also. Typically, a pigment to binder weight ration of about 10/100 to 300/100 is acceptable. The pigments are typically ground using conventional dispersion equipment such as sand mills, pearl mills, ball mills, horizontal mills, and vertical mills. Optional wetting agents, surfactants, and dispersing aids can be employed. This component may also contain a wax composed of ethylene and/or ethylenically unsaturated monomers alkyl (meth)acrylates such as methyl (meth)acrylate, ethyl (meth)acrylate, propyl (meth)acrylate, 2-propyl (meth)acrylate, butyl (meth)acrylate, 2-butyl(meth)acrylate, 2-methylpropyl (meth)acrylate, t-butyl (meth)acrylate, 2-ethylhexyl (meth)acrylate, cyclohexyl (meth)acrylate, lauryl (meth)acrylate, octyl (meth)acrylate, decyl (meth)acrylate, stearyl (meth)acrylate, dicyclopentadienyl (meth)acrylate, and isobornyl (meth)acrylate; aryl monomers such as styrene, alpha-methyl styrene, vinyl toluene, t-butyl styrene, benzyl (meth)acrylate, and 2-benzylethyl (meth)acrylate; hydroxy monomers such as hydroxyethyl (meth)acrylate, hydroxypropyl (meth)acrylate, hydroxybutyl (meth)acrylate, sec-hydroxybutyl (meth)acrylate, and hydroxyoctyl (meth)acrylate; acid functional monomers such as acrylic acid, methacrylic acid, maleic acid, and itaconic acid. Typically, the particle size of the wax particle in the final dispersion is 0.5 microns to 50 microns, preferably 1.0 to 20.0 microns. The wax is dispersed by conventional techniques or emulsified in to the resin system to the desired particle size. The proper amount of the wax dispersion is then added to the pigmented base. The pH of the pigmented base can be adjusted using ammonia or any water soluble primary amines, secondary amines, tertiary amines, polyamines and hydroxyamines, such as ethanolamine, diethanolamine, triethanolamine, n-methylethanolamine, N,N diethylethanolamine, N-aminoethanolamine, N-methyldiethanolamine, monoisopropanolamine, diisopropanolamine, triisopropanolamine, hydroxyamine, butanolamine, hexanolamine, methyl diethanolamine, N,N-diethylaminoethylamine, ethylene diamine, diethylene triamine, diethylene tetraamine, hexamethylene tetramine, triethylamine and the like. The preferred base is ammonia.

#### DESCRIPTION OF INVENTION

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This invention is directed to a coating composition for a water borne base coat especially designed for the automotive market. This invention describes a method for the introduction of metallic flakes into a water borne coating resulting in a highly superior base coat. The proper metal fixation is accomplished by the use of a novel rheology control agent which employs a synthetic sodium lithium magnesium silicate hectorite clay and a polypropylene glycol. The metallic base coat is broken down into three bases that are combined just prior to application of the base coat. The nonmetallic base coat is broken down into two bases that are combined just prior to application of the base coat.

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The invention describes a method for the storage of aluminum flake in an organic medium with introduction of the flakes directly into the aqueous environment just prior to application of the coating system. An acrylic resin with both nonionic and acid functionality has been developed that allows for this introduction of the aluminum flakes into the aqueous environment without the neutralization step. This method minimizes the potential hazard of the aluminum flake reacting with water to form hydrogen gas during storage.

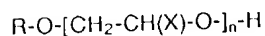
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Component A, aluminum base, is an aluminum slurry in an organic environment containing an acrylic resin having nonionic and acid, preferably nonionic urethane and acid functionality. The aluminum dispersion resin is an acrylic polymer in water-miscible solvent(s), formed by the free-radical initiated polymerization of ethylenically unsaturated monomers. A polyether containing ethylenically unsaturated monomer, preferably a polyether-urethane containing ethylenically unsaturated monomer is included in the monomer mixture, to give a final poly(ethylene oxide) level of 1-50% by weight, preferably 5-20% by weight. The number average molecular weight of the polymer is between 1,000 and 50,000 atomic mass units, its Glass Transition Temperature (Tg) is between -20° and +50° C (calculated by the Fox method), its hydroxyl number is 0 to 50, and its acid number is from 5 to 50 milligrams of KOH per gram of resin solids. The viscosity of the final polymer is 3,000-10,000 cps with a solids content of 60-70% by weight at 23° C.

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Examples of polyether containing monomers include poly(ethylene glykol)mono(meth)acrylate (containing 5 to 20 ethylene oxide units) and any other ethylenically unsaturated monomer that contains polyether units.

The ethylenically unsaturated urethane-polyether containing monomer can be synthesized by reacting a capped polyether glycol with a multifunctional isocyanate in such a way where the final adduct has one remaining free isocyanate. The capped polyether glycol has the following general formula:



wherein

X represents a monofunctional radical having from 1 to 8 carbon atoms, preferably a hydrogen or a methyl group.

R represents a monofunctional, difunctional, trifunctional hydrocarbon radicals having 1 to 40 carbons,

represents the number of moles of ethylene oxide and can range from 0 to 50.

Said capped polyether glycols are known in the art, including polyethylene glycol monomethyl ether, polyethylene glycol monoethyl ether, polyethylene glycol monopropyl ether, polyethylene glycol monobutyl ether, polyethylene glycol nonylphenol ether, polypropylene glycol monomethyl ether, polypropylene glycol monoethyl ether, polypropylene glycol monopropyl ether, and polypropylene glycol monobutyl ether.

The multifunctional isocyanate has the following general formula:



wherein

Z represents a hydrocarbon radical having from 1 to 18 carbon and nitrogen atoms.

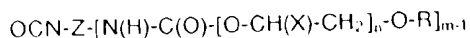
m represents the number of isocyanate groups.

The organic polyisocyanate which is reacted with the polyether glycol as described is essentially any polyisocyanate and is preferably a diisocyanate, e.g., hydrocarbon diisocyanates or substituted hydrocarbon diisocyanates. Many such organic diisocyanates are known in the art, including p-phenylene diisocyanate, biphenyl 4,4'-diisocyanate, toluene diisocyanate, 3,3'-dimethyl-4,4'-biphenylene diisocyanate, 1,4-tetramethylene diisocyanate, 1,6-hexamethylene diisocyanate, 2,2,4-trimethylhexane-1,6 diisocyanate, methylene bis (phenyl isocyanate), 1,5 naphthalene diisocyanate, bis (isocyanatoethylfumarate), isophorone diisocyanate (IPDI) and methylene-bis-(4-cyclohexylisocyanate). There can also be employed isocyanate terminated adducts of polyols such as ethylene glycol, or 1,4-butylene glycol, trimethylolpropane etc. These are formed by reacting more than one mole of a diisocyanate, such as those mentioned, with one mole of polyol to form a longer chain diisocyanate.

It is preferred to employ an aliphatic diisocyanate, since it has been found that these provide better color stability in the finished coating. Examples include 1,6-hexamethylene diisocyanate, 1,4-butylene diisocyanate, methylene bis (4-cyclohexyl isocyanate) and isophorone diisocyanate. Mixtures of diisocyanates can also be employed.

Other multifunctional isocyanates can also be employed such as 1,2,4-benzenetriisocyanate, poly-methylene polyphenyl isocyanate, isocyanurate of isophorone diisocyanate, isocyanurate of hexamethylene diisocyanate.

The adduct may be prepared in conventional manner by reacting a stoichiometric excess of the organic polyisocyanate compound with the appropriate polyether glycol under anhydrous conditions at a temperature between about 20 C to 130 C until the reaction between the isocyanate groups and the active hydrogen group of the polyether glycol is complete. The isocyanate groups and the active hydrogen group of the polyether glycol are allowed to react in such a way that the stoichiometry of the isocyanate group to the active hydrogen groups is 1.5:1 to about 3:1, preferably 2:1. The isocyanate-polyether adduct that is formed has the following general formula:



wherein:

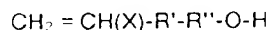
X represents a monofunctional radical having from 1 to 8 carbon atoms, preferably a hydrogen or a methyl group.

R represents a monofunctional, difunctional, trifunctional hydrocarbon radicals having 1 to 40 carbons,

- n represents the number of moles of ethylene oxide and can range from 0 to 50.  
 Z represents a hydrocarbon radical having from 1 to 18 carbon and nitrogen atoms,  
 m represents the number of isocyanate groups.

The reaction between the isocyanate group and the active hydrogen group of the polyether glycol may also be catalyzed. Examples of such catalyst are dibutyltin dilaurate, dibutyltin diacetate, 1,4-diaza(2,2,2)-bicyclooctane, tetramethylbutanediamine, zinc octoate and stannous octate. Solvents used must be nonreactive toward the isocyanate and also be water miscible. Examples of solvents meeting these requirements are acetone, methylethylketone, dimethylformamide, N-methylpyrrolidone, ethylene glycol diacetate and propylene glycol diacetate.

The urethane-polyether monomer is prepared in conventional manner by reacting an equal stoichiometric amount of an ethylenically unsaturated compound having active hydrogen groups. The general formula for this ethylenically unsaturated compound is as follows:



wherein:

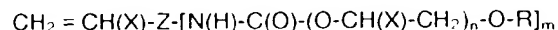
X represents a monofunctional radical having from 1 to 8 carbon atoms, preferably a hydrogen or methyl group,

R' represents a radical being a carboxyl, aliphatic, or aryl, preferably carboxyl.

R'' represents a hydrocarbon radical having from 1 to 40 carbon atoms.

Examples of hydroxy functional ethylenically unsaturated monomers are hydroxyethyl (meth)acrylate, hydroxypropyl (meth)acrylate, hydroxybutyl (meth)acrylate, sec-hydroxybutyl (meth)acrylate, and hydroxyoctyl (meth)acrylate.

The final urethane-polyether monomer has the following general structure:



wherein:

X represents a monofunctional radical having from 1 to 8 carbon atoms, preferably a hydrogen or methyl group,

R' represents a radical being a carboxyl, aliphatic, or aryl, preferably carboxyl.

R'' represents a hydrocarbon radical having from 1 to 40 carbon atoms.

n represents the number of moles of ethylene oxide and can range from 0 to 50.

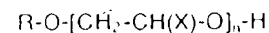
Z represents a hydrocarbon radical having from 1 to 18 carbon and nitrogen atoms,

m represents the number of isocyanate groups.

The urethane-polyether monomer must be prepared under anhydrous conditions at a temperature between about 20°C to 130°C until the reaction between the isocyanate and the urethane-polyether adduct is complete. Any residual isocyanate may be destroyed by reaction with a mono functional active compound. Examples of such compounds are methanol, ethanol, propanol, iso-propanol, butanol and hexanol. The reaction between the isocyanate and the active hydrogen compound may also be catalyzed. Examples of catalysis are dibutyltin dilaurate, dibutyltin diacetate, 1,4-diaza(2,2,2)bicyclooctane, tetramethylbutanediamine, zinc octoate

and stannous octoate. Solvents used for this reaction must be anhydrous and nonreactive toward the isocyanate group. Examples of such solvents are acetone, methylethyl ketone, dimethyl formamide, N-methylpyrrolidone, ethylene glycol diacetate, and propylene glycol diacetate. A free radical inhibitor can also be present in the reaction mixture. The preferred inhibitor is butylated hydroxy toluene.

The ethylenically unsaturated polyether-urethane containing monomer also can be synthesized by reacting a capped polyether glycol with a radical having isocyanate and ethlenically unsaturated functionality in such a way where the final urethane-polyether monomer has ethlenically unsaturated functionality for free radical polymerization. The capped polyether glycol has the following general formula:



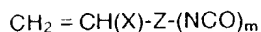
wherein:

X represents a monofunctional radical having from 1 to 8 carbon atoms, preferably a hydrogen or a methyl group.

R represents a monofunctional, difunctional, trifunctional hydrocarbon radicals having 1 to 40 carbons,

n represents the number of moles of ethylene oxide and can range from 0 to 50.

The multifunctional isocyanate has the following general formula:



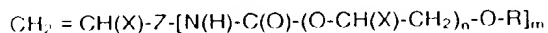
wherein

Z represents a hydrocarbon radical having from 1 to 18 carbon and nitrogen atoms,

m represents the number of isocyanate groups.

X represents a monofunctional radical having from 1 to 8 carbon atoms, preferably a hydrogen or a methyl group.

The organic polyisocyanate which is reacted with the polyether glycol as described is essentially any polyisocyanate and is preferably a monoisocyanate with ethylenically unsaturated functionality. The preferred isocyanate compound having ethylenically unsaturated functionality, is meta-isopropenyl-alpha, alpha-dimethylbenzylisocyanate. The urethane-polyether monomer may be prepared in conventional manner by reacting a stoichiometric excess of the organic polyisocyanate compound with the appropriate polyether glycol under anhydrous conditions at a temperature between about 20°C to 130°C until the reaction between the isocyanate groups and the active hydrogen group of the polyether glycol is complete. The isocyanate groups and the active hydrogen group of the polyether glycol are allowed to react in such a way that the stoichiometry of the isocyanate group to the active hydrogen groups is 1.0:0.8 to about 1:1, preferably 1:1. The general formula for this urethane-polyether monomer is as follows:



wherein:

X represents a monofunctional radical having from 1 to 8 carbon atoms, preferably a hydrogen or a methyl group.

R represents a monofunctional, difunctional, trifunctional hydrocarbon radicals having 1 to 40 carbons,

n represents the number of moles of ethylene oxide and can range from 0 to 50.

Z represents a hydrocarbon radical having from 1 to 18 carbon and nitrogen atoms,

m represents the number of isocyanate groups.

The reaction between the isocyanate group and the active hydrogen group of the polyether glycol may also be catalyzed. Examples of such catalyst are dibutyltin dilaurate, dibutyltin diacetate, stannous octate, 1,4-diaza(2,2,2)bicyclooctane, tetramethylbutanediamine, stannous and zinc octoate. Solvents used must be nonreactive toward the isocyanate and also be water miscible. Examples of solvents meeting these requirements are acetone, methylethylketone, dimethylformamide, N-methylpyrrolidone, ethylene glycol diacetate and propylene glycol diacetate. A free radical inhibitor can also be present in the reaction mixture. The preferred inhibitor is butylated hydroxy toluene.

The final aluminum dispersion resin is an acrylic polymer in water miscible solvent(s), formed by the free radical initiated polymerization of ethylenically unsaturated monomers including the polyether containing monomer and/or the urethane-polyether monomer previously described.

Examples of ethylenically unsaturated monomers which can be used for this purpose include alkyl (meth)acrylates such as methyl (meth)acrylate, ethyl (meth)acrylate, propyl (meth)acrylate, 2-propyl (meth)acrylate, butyl (meth)acrylate, 2-butyl (meth)acrylate, 2-methylpropyl (meth)acrylate, t-butyl (meth)acrylate, 2-ethylhexyl (meth)acrylate, cyclohexyl (meth)acrylate, lauryl (meth)acrylate, octyl (meth)acrylate, decyl (meth)acrylate, stearyl (meth)acrylate, dicyclopentadienyl (meth)acrylate, and isobornyl (meth)acrylate; aryl monomers such as styrene, alpha-methyl styrene, vinyl toluene, t-butyl styrene, benzyl (meth)acrylate, and 2-benzylethyl (meth)acrylate; hydroxy monomers such as hydroxyethyl (meth)acrylate, hydroxypropyl (meth)acrylate, allyl alcohol, hydroxybutyl (meth)acrylate, sec-hydroxybutyl (meth)acrylate, and hydroxyoctyl (meth)acrylate; acid functional monomers such as acrylic acid, methacrylic acid, maleic acid, and itaconic acid.

The acrylic polymer is prepared in organic solvents that are water miscible, such as methyl ethyl ketone, acetone, ethanol, methanol, propanol, butanol, N-methylpyrrolidone, glycols, glycol ethers, glycol acetates, diethylene glycol ethers, diethylene glycol acetates, propylene glycol ethers, propylene glycol acetates, dipropylene glycol ethers, dipropylene glycol acetates, specially preferred are ethylene glycol propyl ether, ethylene glycol butyl ether, ethylene glycol hexyl ether, propylene methyl ether, propylene ethyl ether, propylene propyl ether, propylene butyl ether, propylene hexyl ether, dipropylene methyl ether, dipropylene ethyl ether, dipropylene propyl ether, methyl ethyl ketone and dipropylene butyl ether. The



acrylic polymer is prepared via normal solution polymerization techniques. Examples of free-radical initiators include peroxides, such as benzoyl peroxide, tertiary-butyl peracetate, tertiary-butyl peroctoate, tertiary-butyl perbenzoate, and ditertiary butyl peroxide; and 1,1'-azobis(alkyl nitrile)s, such as , 'azobis(isobutyronitrile), and 1,1'-azobis(cyclohexane carbonitrile). Optionally, surfactants can also be present during the polymerization at a level of 0.1-2.0% by weight.

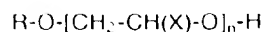
The novel water reducible acrylic resin having nonionic, urethane and acid functionality also can be synthesized by the free-radical initiated polymerisation of  $\alpha,\beta$ -ethylenically unsaturated monomers, to form an acrylic polymer, having active hydrogen functionality followed by grafting of a diisocyanate / polyether adduct onto the acrylic polymer, to give a final poly(ethylene oxide) level of 1-50 % by weight, preferable 5 % to 20 % by weight. The number average molecular weight of the polymer is between 1,000 and 50,000, preferably between 1,000 and 20,000 atomic mass units, its Glass Transition Temperature ( $T_g$ ) is between  $-20^\circ$  and  $+50^\circ$  C (calculated by the Fox method), its hydroxyl number 0 to 50 and its acid number is from 5 to 50, preferable from 10 to 20 milligrams of KOH per gram of resin solids. The viscosity of this polymer is between 5,000 and 15,000 cps at non-volatile by weight, measured at  $25^\circ$  C.

The active hydrogen functional acrylic polymer is synthesized in water miscible solvent(s), formed by the free radical initiated polymerization of ethylenically unsaturated monomers which include active hydrogen and acid functionality. The final acrylic resin should have a number average molecular weight of the polymer is between 1,000 and 20,000 atomic mass units, its Glass Transition Temperature ( $T_g$ ) is between  $-20^\circ$  and  $+50^\circ$  C (calculated by the Fox method), its hydroxyl number is from 0 to 75 and its acid number is from 10 to 20 milligrams of KOH per gram of resin solids. The viscosity of this polymer is between 5,000 and 15,000 cps at 66% non-volatile by weight, measured at  $25^\circ$  C.

Examples of ethylenically unsaturated monomers which can be used for this purpose include alkyl (meth)acrylates such as methyl (meth)acrylate, ethyl (meth)acrylate, propyl (meth)acrylate, 2-propyl (meth)acrylate, butyl (meth)acrylate, 2-butyl (meth)acrylate, 2-methylpropyl (meth)acrylate, t-butyl (meth)acrylate, 2-ethylhexyl (meth)acrylate, cyclohexyl (meth)acrylate, lauryl (meth)acrylate, octyl (meth)acrylate, decyl (meth)acrylate, stearyl (meth)acrylate, dicyclopentadienyl (meth)acrylate, and isobornyl (meth)acrylate; aryl monomers such as styrene, alpha-methyl styrene, vinyl toluene, t-butyl styrene, benzyl (meth)acrylate, and 2-benzylethyl (meth)acrylate; ethylenically unsaturated compounds having at least one active hydrogen group such as, hydroxy monomers such as hydroxyethyl (meth)acrylate, hydroxypropyl (meth)acrylate, hydroxybutyl (meth)acrylate, sec-hydroxybutyl (meth)acrylate, and hydroxyoctyl (meth)acrylate; acid functional monomers such as acrylic acid, methacrylic acid, maleic acid, and itaconic acid.

The acrylic polymer is prepared in organic solvents that are water miscible, such as methyl ethyl ketone, acetone, ethanol, methanol, propanol, butanol, N-methylpyrrolidone, glycols, glycol ethers, glycol acetates, diethylene glycol ethers, diethylene glycol acetates, propylene glycol ethers, propylene glycol acetates, dipropylene glycol ethers, dipropylene glycol acetates, specially preferred are ethylene glycol propyl ether, ethylene glycol butyl ether, ethylene glycol hexyl ether, propylene methyl ether, propylene ethyl ether, propylene propyl ether, propylene butyl ether, propylene hexyl ether, dipropylene methyl ether, dipropylene ethyl ether, dipropylene propyl ether, methyl ethyl ketone and dipropylene butyl ether. The acrylic polymer is prepared via normal solution polymerization techniques. Examples of free-radical initiators include peroxides, such as benzoyl peroxide, tertiary-butyl peracetate, tertiary-butyl peroctoate, tertiary-butyl perbenzoate, and ditertiary butyl peroxide; and 1,1'-azobis(alkyl nitrile)s, such as , 'azobis(isobutyronitrile), and 1,1'-azobis(cyclohexane carbonitrile). Optionally, surfactants can also be present during the polymerization at a level of 0.1-2.0% by weight.

The monoisocyanate:polyether urethane adduct is synthesized by reacting a capped polyether glycol with a multifunctional isocyanate in such a way where the final adduct has one remaining free isocyanate. The capped polyether glycol has the following general formula:



wherein

X represents a monofunctional radical having from 1 to 8 carbon atoms, preferably a hydrogen or a methyl group

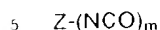
R represents a monofunctional, difunctional, trifunctional hydrocarbon radicals having 1 to 40 carbons,

n represents the number of moles of ethylene oxide and can range from 0 to 50.

The capped polyether has a molecular range of 100 to 4,000. Examples of capped polyether which can be used include polyethylene glycol monomethyl ether, polyethylene glycol monethyl ether, polyethylene glycol monopropyl ether, polyethylene glycol monobutyl ether, polyethylene glycol nonylphenol ether,

polypropylene glycol monomethyl ether, polypropylene glycol monoethyl ether, polypropylene glycol monopropyl ether, and polypropylene glycol monobutyl ether.

The multifunctional isocyanate has the following general formula:



wherein:

Z represents a hydrocarbon radical having from 1 to 18 carbon and nitrogen atoms,

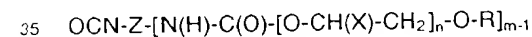
m represents the number of isocyanate groups.

10 The organic polyisocyanate which is reacted with the polyether glycol as described is essentially any polyisocyanate and is preferably a diisocyanate, e.g., hydrocarbon diisocyanates or substituted hydrocarbon diisocyanates. Many such organic diisocyanates are known in the art, including p-phenylene diisocyanate, biphenyl 4,4'-diisocyanate, toluene diisocyanate, 3,3'-dimethyl-4,4'-biphenylene diisocyanate, 1,4-tetramethylene diisocyanate, 1,6-hexamethylene diisocyanate, 2,2,4-trimethylhexane-1,6 diisocyanate, 15 methylene bis (phenyl isocyanate), 1,5 naphthalene diisocyanate, bis(isocyanatoethylfumarate), isophorone diisocyanate (IPDI) and methylene-bis-(4-cyclohexylisocyanate). There can also be employed isocyanate terminated adducts of polyols such as ethylene glycol, or 1,4-butylene glycol, trimethylolpropane etc. These are formed by reacting more than one mole of a diisocyanate, such as those mentioned, with one mole of polyol to form a longer chain diisocyanate.

20 It is preferred to employ an aliphatic diisocyanate, since it has been found that these provide better color stability in the finished coating. Examples include 1,6-hexamethylene diisocyanate, 1,4-butylene diisocyanate, methylene bis (4-cyclohexyl isocyanate) and isophorone diisocyanate. Mixtures of diisocyanates can also be employed.

Other multifunctional isocyanates can also be employed such as 1,2,4-benzenetriisocyanate, poly- 25 methylene polyphenyl isocyanate, isocyanurate of isophorone diisocyanate, isocyanurate of hexamethylene diisocyanate.

The adduct may be prepared in conventional manner by reacting a stoichiometric excess of the organic polyisocyanate compound with the appropriate polyether glycol under anhydrous conditions at a temperature between about 20 °C to 130 °C until the reaction between the isocyanate groups and the active hydrogen 30 group of the polyether glycol is complete. The isocyanate groups and the active hydrogen group of the polyether glycol are allowed to react in such a way that the stoichiometry of the isocyanate group to the active hydrogen groups is 1.5:1 to about 3:1, preferably 2:1. the isocyanate-polyether adduct that is formed has the following general formula:



wherein:

X represents a monofunctional radical having from 1 to 8 carbon atoms, preferably a hydrogen or a methyl group.

40 R represents a monofunctional, difunctional, trifunctional hydrocarbon radicals having 1 to 40 carbons,

n represents the number of moles of ethylene oxide and can range from 0 to 50.

Z represents a hydrocarbon radical having from 1 to 18 carbon and nitrogen atoms,

m represents the number of isocyanate groups.

45 The reaction between the isocyanate group and the active hydrogen group of the polyethylene glycol may also be catalyzed. Examples of such catalyst are dibutyltin dilaurate, dibutyltin diacetate, 1,4-diaza-(2,2,2)bicyclooctane, tetramethylbutanediamine, zinc octoate and stannous octate. Solvents used must be nonreactive toward the isocyanate and also be water miscible. Examples of solvents meeting these requirements are acetone, methylethylketone, dimethylformamide, N-methylpyrrolidone, ethylene glycol 50 diacetate and propylene glycol diacetate.

The final water reducible resin having nonionic, urethane and acid functionality is prepared by reacting the monoisocyanate/polyether urethane adduct with the active hydrogen functional acrylic polymer in a conventional manner under anhydrous conditions at a temperature between 20 °C to 130 °C until the reaction between the isocyanate groups and the active hydrogen group of the acrylic is complete. The stoichiometry 55 of the isocyanate group of the adduct to the active hydrogen groups of the acrylic resin is 0.1:1.0 to 1.0:1.0. The final polymer should have a final poly(ethylene oxide) level of 1 to 50 % by weight, preferably 5 % to 20 % by weight. The number average molecular weight of the polymer is between 1,000 and 50,000 atomic mass units, its Glass Transition Temperature (Tg) is between -20 °C and +50 °C (calculated by the Fox

method), its hydroxyl number is from 0 to 50 and its acid number is from 5 to 50, preferably 10 to 20 milligrams of KOH per gram of resin solids. The viscosity of this polymer is between 5,000 and 15,000 cps at 66 % nonvolatile by weight, measured at 25 °C.

The reaction between the isocyanate group and the active hydrogen group of the polyethylene glycol may also be catalyzed. Examples of such catalyst are dibutyltin dilaurate, dibutyltin diacetate, 1,4-diaza-(2,2,2)bicyclooctane, tetramethylbutanediamine, zinc octoate and stannous octate. Solvents used must be nonreactive toward the isocyanate and also be water miscible. Examples of solvents meeting these requirements are acetone, methylethylketone, dimethylformamide, N-methylpyrrolidone, ethylene glycol diacetate and propylene glycol diacetate.

Any commercial aluminum paste available for solvent borne or water borne applications can be used in the preparation of component A. The amount of aluminum flake in the component is 4-30% by weight, preferably 23-30% by weight.

Other nonmetallic and metallic pigments that can be incorporated into this base include copper, copper alloys, mica, coated metallic foils, plastic flakes and steel. However, this is not an inclusive list.

Additional organic solvent may be needed to reduce the viscosity of the acrylic resin. Acceptable solvents must be water miscible such as methyl ethyl ketone, acetone, ethanol, methanol, propanol, butanol, N-methylpyrrolidone, glycols, glycol ethers, glycol acetates, diethylene glycol ethers, diethylene glycol acetates, propylene glycol ethers, propylene glycol acetates, dipropylene glycol ethers, dipropylene glycol acetates, specially preferred are ethylene glycol propyl ether, ethylene glycol butyl ether, ethylene glycol hexyl ether, propylene methyl ether, propylene ethyl ether, propylene propyl ether, propylene butyl ether, propylene hexyl ether, dipropylene methyl ether, dipropylene ethyl ether, dipropylene propyl ether and dipropylene butyl ether. Component A, the aluminum base, has a total solids level of 25-55 % by weight, 5-35 %, preferably 20-30 %, binder by weight, 35-50% organic solvent by weight and total metallic pigment level of 4-30% by weight. Additional wetting, anti-settling additives common to the paint industry may be added.

The preparation of component A, the aluminum base, is critical for the proper appearance of the aluminum flake in the final coating. The proper amount of the acrylic resin solution is dissolved into the proper amount of water miscible organic solvent, such as methyl ethyl ketone, acetone, ethanol, methanol, propanol, butanol, N-methylpyrrolidone, glycols, glycol ethers, glycol acetates, diethylene glycol ethers, diethylene glycol acetates, propylene glycol ethers, propylene glycol acetates, dipropylene glycol ethers, dipropylene glycol acetates, specially preferred are ethylene glycol propyl ether, ethylene glycol butyl ether, ethylene glycol hexyl ether, propylene methyl ether, propylene ethyl ether, propylene propyl ether, propylene butyl ether, propylene hexyl ether, dipropylene methyl ether, dipropylene ethyl ether, dipropylene propyl ether and dipropylene butyl ether. The proper amount of the aluminum paste is slurred into the above described resin solution. The aluminum slurry must be stirred until the base is smooth and uniform.

Component B, pigmented base, imparts the coloring effect upon the coating composition. The resin used for the dispersing of the pigments is an acrylic latex. Typical properties of this acrylic latex are as follows:

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	Solids content, %	20-60
5	pH	7 to 10
	Weight per gallon, lbs	8.0 to 9.0
10	Minimum film formation temperature, C	0 to 70
	Sward hardness	25 to 110
	Mechanical stability	OK
15	Glass transition temperature, C	10 to 50
	Average molecular weight	150,000 to
20	300,000	
	Number average molecular weight	30,000 to 60,000
	Acid number on solids	5 to 35
25	Volatile organic solvents, %	0 to 20
30	Particle size, nm	50 to 150

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Especially preferred acrylic lattices are Neocryl A-622, Neocryl A-640, Neocryl A-6037 sold by ICI Resins, Joncryl J-537, Joncryl J-538 sold by Johnson Wax, Aralon 860-W-45 sold by NL Industries. The level of this resin in component C is 15-60% by weight, preferably 20-40% by weight. The organic, inorganic pigments and/or dyestuffs are ground with this resin employing standard techniques. Examples of dispersing equipment are ball mills, pebble mills, pearl mills, horizontal mills, and vertical mills. Examples of pigments and dyestuffs but not limited to are titanium dioxide, graphite, carbon black, zinc oxide, cadmium sulfide, chromium oxide, zinc sulfide, zinc chromate, strontium chromate, barium chromate, lead chromate lead cyanamide, lead silico chromate, chromium oxide, zinc sulfide, yellow nickel titanium, yellow chromium titanium, red iron oxide, yellow iron oxide, transparent red iron oxide, transparent yellow oxide, black iron oxide, naphthol reds and browns, anthraquinones, dioxazine violet, isoindoline yellows, arylide yellow and oranges, ultramarine blue, phthalocyanine complexes, amaranth, quinacridones, halogenated thioindigo pigments and extender pigments such as magnesium silicate, aluminum silicate, calcium silicate, calcium carbonate, fumed silica, barium sulfate, among others. The level of pigment/dyestuff in component C is 0.0-45.0% by weight, typically 0-25%. This component may also contain a wax composed of ethylene and/or ethylenically unsaturated monomers alkyl (meth)acrylates such as methyl (meth)acrylate, ethyl (meth)acrylate, propyl (meth)acrylate, 2-propyl (meth)acrylate, butyl (meth)acrylate, 2-butyl(meth) acrylate, 2-methylpropyl (meth)acrylate, t-butyl (meth)acrylate, 2-ethylhexyl (meth)acrylate, cyclohexyl (meth)acrylate, lauryl (meth)acrylate, octyl (meth)acrylate, decyl (meth)acrylate, stearyl (meth)acrylate, dicyclopentadienyl (meth)acrylate, and isobornyl (meth)acrylate; aryl monomers such as styrene, alpha-methyl styrene, vinyl toluene, t-butyl styrene, benzyl (meth)acrylate, and 2-benzylethyl (meth)acrylate, hydroxy monomers such as hydroxyethyl (meth)acrylate, hydroxypropyl (meth)acrylate, hydroxybutyl (meth)acrylate, sec-hydroxybutyl (meth)acrylate, and hydroxyoctyl (meth)acrylate, acid functional monomers such as acrylic acid, methacrylic acid, maleic acid, and itaconic acid. Typically, the particle size of the wax particle in the final

dispersion is 0.5 micron to 50 micron, preferably 1.0 micron to 2.0 micron. Typical the wax contain 0 - 20 % by weight vinyl acetate. Typical properties of this wax are as follows:

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5	Acid Number (mg KOH/g)	0-50
	Hardness, dmm (ASTM D-5)	<2.0
10	Mettler Drop Point, C (ASTM D-3104)	100-150
	Particle Size, micron	0.5-10.0
15	Hydroxyl Number	0-50

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Typically, the wax is dispersed by conventional techniques in to the resin system to the desired particle size or be prepared by emulsion techniques. The proper amount of the wax dispersion is then added to the pigmented base. This component may also contain other conventional paint additives such as dispersing aids, anti-settling aids, wetting aids, thickening agents, extenders, plasticizers, stabilizers, light stabilizers, antifoams, defoamers and catalysts singly or a multitude of them at the conventional levels. Also present in this component is a water miscible solvent such as methyl ethyl ketone, acetone, ethanol, methanol, propanol, butanol, N-methylpyrrolidone, glycols, glycol ethers, glycol acetates, diethylene glycol ethers, diethylene glycol acetates, propylene glycol ethers, propylene glycol acetates, dipropylene glycol ethers, dipropylene glycol acetates, specially preferred are ethylene glycol propyl ether, ethylene glycol butyl ether, ethylene glycol hexyl ether, propylene methyl ether, propylene ethyl ether, propylene propyl ether, propylene butyl ether, propylene hexyl ether, dipropylene methyl ether, dipropylene ethyl ether, dipropylene propyl ether and dipropylene butyl ether at a level of 0-25% by weight, preferably 0-12% by weight. Examples of water reducible polyurethanes are NeoRez R-960, R-966, R-967, R-9637 manufactured by ICI Resins and Spensol L51 and L53 manufactured by NL Industries.

Component B, the pigmented base, has a total solids level of 5-60% by weight, a total pigment level of 0.0-45% by weight, a total organic solvent level of 0-25% by weight, a total binder of 20-60% by weight, a total wax level of 0-20% by weight, a total polyurethane level of 0-25% by weight, a total water level of 40 - 95 % by weight and a pH range of 7 - 10, preferably 7 - 9. The pH of the base can be adjusted using ammonia or other water soluble primary amines, secondary amines, tertiary amines, polyamines and hydroxyamines, such as ethanolamine, diethanolamine, triethanolamine, N-methylethanolamine, N,N-diethylethanolamine, N-aminoethanolamine, N-methyldiethanolamine, monoisopropanolamine, diisopropanolamine, triisopropanolamine, hydroxyamine, butanolamine, hexanolamine, methyl diethanolamine, N,N-diethylaminoethylamine, ethylene diamine, diethylene triamine, diethylene tetraamine, hexamethylene tetramine, triethylamine and the like.

Component C, reducer base, is an aqueous solution containing a dispersion of synthetic sodium lithium magnesium silicate hectorite clay, a polypropylene glycol and/or an anionic surfactant and optionally a water reducible polyurethane resin. Addition of a polypropylene glycol to the hectorite clay results in a low viscosity dispersion. The low viscosity dispersion of the hectorite clay allows for easy incorporation of the dispersion into the paint composition and allows the coating to be applied wetter during the application process. The polypropylene glycol also allows the coating to be sprayed wetter onto the substrate. Addition of the anionic surfactant results in the pregelling of the hectorite clay. Using the pregelled hectorite clay in a coating composition results in a drier film application and a stable viscosity. The typical properties of the synthetic sodium lithium magnesium silicate clay are as follows:

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	Solids Content, %	95-100
5	Typical Analysis	
	SiO <sub>2</sub>	50-65
	MgO	25-35
10	Li <sub>2</sub> O	0.1-1.5
	Na <sub>2</sub> O	1.5-3.5
15		3.0-12.0
20	Appearance	White free flowing powder
	Particle Size	<2% greater than 250 microns
25	Surface Area, m <sup>2</sup> /g	370s

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Especially preferred synthetic sodium lithium magnesium silicate hectorite clay is Laponite RD, available from Laporte, Inc. This hectorite clay is present in this component at a level of 0.1-10% by weight, preferably 0.5-3.0% by weight. This component also contains a polypropylene glycol at a level of 0.1 - 10 % by weight and/or an anionic surfactant at a level of 0.1 - 2.0 % by weight, preferably 0.1 - 1.0 % by weight. The molecular weight range of the polypropylene glycol is 435 to 3900. Examples of anionic surfactants that can be employed are alcohol sulfated, ether sulfated, linear alkyl benzene sulfonates and specially preferred dialkyl sulfosuccinates. The cation of the salt may be ammonium, sodium, potassium, and other monovalent and/or divalent metals. This component may also contain a water reducible polyurethane at a level of 0.0 - 25.0 % by weight, preferably 0.0 - 10.0 % by weight. Examples of water reducible polyurethanes are NeoRez R-960, R-966, R-967, R-9637 manufactured by ICI Resins and Spensol L51 and L53 manufactured by NL Industries.

The hectorite clay can be dispersed in the aqueous medium by Cowles or agitation. A water miscible solvent may also be present in this component such as methyl ethyl ketone, acetone, ethanol, methanol, propanol, butanol, N-methylpyrrolidone, glycols, glycol ethers, glycol acetates, diethylene glycol ethers, diethylene glycol acetates, propylene glycol ethers, propylene glycol acetates, dipropylene glycol ethers, dipropylene glycol acetates, specially preferred are ethylene glycol propyl ether, ethylene glycol butyl ether, ethylene glycol hexyl ether, propylene methyl ether, propylene ethyl ether, propylene propyl ether, propylene butyl ether, propylene hexyl ether, dipropylene methyl ether, dipropylene ethyl ether, dipropylene propyl ether and dipropylene butyl ether at a level of 0-25% by weight, preferably 0-12% by weight. The total water content of this component is 50 - 99.9 % by weight.

Other water reducible resin may be added to the coating to improve certain film properties such as aqueous one-component aliphatic polyurethane dispersions, aromatic polyurethane dispersions, polyesters, and alkyds.

Addition of an acrylic acid and/or polyurethane based thickeners known to the art may also be added to achieve desired properties. The incorporation of metal oxides such as zinc oxide may also improve the humidity resistance of the final film.

Additional anionic and nonionic surfactants can be added to the paint composition to increase wetting of the substrate by the coating such as FC-120, FC-430 sold by 3M, Surfynol 104, Surfynol 440 sold by Air

Products, Triton X-100 sold by Rohm and Haas, Troysol LAC sold by Troy Chemical Company, Aerosol OT's sold by American Cyanamid including the salts of dialkyl sulfosuccinates and Igepal's sold by GAF including the ethoxylated alkyl phenols.

This base coat composition exhibits a very high viscosity at low shear rate while having a low viscosity at high shear rates. This phenomenon results in the outstanding metal effect this coating composition possesses. The efflux time of the coating through a #4 Ford cup is 15 to 30 seconds at 23 °C.

Using the component B and component C nonmetallic automotive base coat kit compositions can be produced.

The coating composition can be applied using conventional spray equipment or high volume low pressure spray equipment resulting in a high quality finish. Other modes of application are roller coating, brushing, sprinkling, flow coating, dipping, electrostatic spraying, or electrophoresis. Suitable substrates may be made of wood, metal, and synthetic material. After a flash off time of 15 to 45 minutes, a transparent top coat can be applied over this base coat. The preferred transparent clear coat is two component, based on acrylic and/or polyester resins cured with polyisocyanates. Other examples of cured transparent clear coats are acrylic and/or polyester resins cured with silanes, acrylic and/or polyester acid cured epoxy coatings, and acrylic and/or polyester resins cured with aminoplast resins, such as melamine resins, however, this is not an inclusive list.

#### PREFERRED EMBODIMENTS OF INVENTION

The following examples illustrate the invention without limiting the scope thereof.

##### Example 1: Base Coat Preparation

In a 1-liter distillation flask, fitted with a water-cooled condenser and a dry air-sparge tube, under a dry air atmosphere, 177.8 grams of isophorone diisocyanate is combined with 422.4 grams of Methoxy Polyethylene Glycol of the formula:  $\text{CH}_3\text{O}-(\text{CH}_2\text{CH}_2\text{O})_n-\text{H}$  (molecular weight = 550), 10 grams of Urethane Grade Methyl Ethyl Ketone, and 12 grams of a 0.5% solution of Butylated Hydroxy Toluene in Urethane Grade Methyl Ethyl Ketone. The mixture is then stirred for 5 minutes at ambient temperature, and 0.2 grams of dibutyltin diacetate is added. The mixture is continuously stirred, and heated as necessary, to maintain a temperature of 60 °C for two hours. The mixture is then cooled to less than 30 °C, and 96.5 grams of Hydroxyethyl Acrylate is added. The mixture is continuously stirred, and heated as necessary, to maintain a temperature of 60 °C for two hours. Any residual isocyanate (as determined by Infrared Spectroscopy) should be reacted with a minimum amount of anhydrous ethanol. The product can be sealed in a dried glass bottle.

Approximately 134 grams of Propylene glycol monopropyl ether is combined with 5 grams of Methyl Ethyl Ketone in a one-liter distillation flask, fitted with a water-cooled condenser, under a Nitrogen blanket. The mixture is brought to reflux (~ 140 °C), the Nitrogen flow reduced to a minimum, and a mixture of  $\alpha,\beta$ -ethylenically unsaturated monomers is pumped into the flask over two hours. The monomer mixture consists of Methyl Methacrylate - 7.3%, Styrene - 18.9%, n-Butyl Methacrylate - 47.9%, Acrylic Acid - 1.6%, 2-Ethylhexyl Acrylate - 2.4%, and Polyether Monomer (described above) - 22.0%. While the monomers are being pumped into the reaction flask, a mixture of 10.6 grams of free-radical initiator (t-Butyl Perbenzoate) and 50 grams solvent (propylene glycol monopropyl ether) is simultaneously fed into the flask over two hours.

After adding all the monomers, solvent, and initiator (and flushing the pumps and feed lines with propylene glycol monopropyl ether), the reaction is held at reflux. After thirty minutes at reflux, a mixture of 1.2 grams of initiator and 35 grams of solvent (propylene glycol monopropyl ether) is pumped into the reaction flask over ten minutes. After adding the initiator and solvent mixture, the reaction is held at reflux. After one to two hours at reflux, the reaction is cooled and filtered into a lined metal can. The viscosity of this polymer is between 2,500 and 4,500 cps at 66.2% non-volatile by weight, measured at 25 °C, and its acid number is from 12.5 to 14.5 milligrams of KOH per gram of resin solids.

##### Preparation of Component A - Aluminum Base

A commercial solvent borne aluminum paste, 60.0g, was added to 60.0g of butoxyethanol. The mixture was stirred until a homogeneous dispersion was produced. To this dispersion, 30.0g, of the resin described above was added to the dispersion and the entire mixture stirred until a smooth, uniform slurry was produced. This aluminum base contains 39% volatile organic solvent by weight, a solids level of 61.0% by

weight, a binder level of 13% by weight and an aluminum flake level of 26.0% by weight.

#### Preparation of Component B - Pigmented Base

5 The following pigment slurry was prepared, 12.2g of NeoCryl A-6037, 14.8g of NeoCryl A-640, 0.3g of 30% aqueous ammonia, 3.2g of Perrindo Maroon R-6424 manufactured by Mobay Corporation, 0.1g of a defoamer and 0.05g of Surfynol 104BC manufactured by Air Products. The above mixture were mixed together and predispersed using a Cowles blade. The mixture was then ground on a sand mill until the desired particle size was achieved. A let down solution containing 29.5g of NeoCryl A-622, 2.6g of butoxyethanol, 1.3g of deionized water and 0.03g of 1% aqueous solution of FC-120 was added to the grind  
10 portion under agitation. An unpigmented base can be prepared using the procedure above with the elimination of the Perrindo Maroon R-6424 pigment. A wax dispersion was prepared by grinding 50.0g of a wax, 50.0g of unpigmented resin solution, 50.0 g of deionized water and 50.0g of butoxyethanol. This paste was ground to the desired particle size using conventional techniques. To the pigmented and unpigmented  
15 bases described above, 6.0g of this wax dispersion was added with stirring. Also added to the pigmented and unpigmented bases described above was 16.0g of NeoRez R-9637 manufactured by ICI Resins. The resulting pigmented base had a pH of 8.4, a solids level of 38.1% by weight, a volatile organic level of 10.5% by weight, a binder level of 34.2% by weight, a pigment level of 4.3% by weight and a water level of 51.4% by weight. The unpigmented base has a solid level of 35.4% by weight, volatile organic level of  
20 9.5% by weight, a water level of 55.1% by weight and a pH of 8.5.

#### Preparation of Component C - Reducer Base

25 To 1953.2g of deionized water, 31.3g of Laponite RD manufactured by Laporte Industries and 10.7g of Pluracol P1010 manufactured by BASF Corporation was added. The resulting solution was then agitated for one hour until the Laponite RD was completely dispersed. To this solution was added 1953.2g of deionized water. The resulting mixture had a Laponite RD level of 0.79% by weight and a polypropylene glycol level of 0.27% by weight.

#### EXAMPLE 2 - PREPARATION OF A WATER BORNE RED METALLIC BASE COAT

Component	Parts by Weight
Component A Aluminum Base	16.0
Component B Pigmented Base	54.0
Component C Reducer Base	85.0

50 The above formula was used to prepare a red metallic base coat. The order of addition is that listed.  
55 After each introduction of a component, the entire solution was stirred. The base coat was sprayed on panels primed with an epoxy primer. The base coat was sprayed on the substrate using conventional spray equipment until hiding was achieved. The base coat was allowed to flash for 30 minutes and then clear coated with a transparent coating based on isocyanate technology. The final properties of the base coat



after coating with a conventual polyurethane clear coat are listed below, including appearance ratings.

Properties of the Red Metallic Base Coat and Base Coat, Clear Coat System

Property	Value
Water	72.5 %
Volatile Organic	7.4 %
Pigment	2.4 %
Nonvolatile	20.1 %
Binder	10.8 %
Weight per Gallon (lbs/gal)	8.4
VOC (lbs/gal)	2.4
Gloss, 20 deg	90
DOI	84
Adhesion	good
Film Thickness (mil)	0.34
Humidity	good
Water Spray	excell.
Metal Control	excell.

### EXAMPLE 3 - Preparation of a Water Borne Silver Metallic Base Coat

Component A	16.0g
Aluminum Base	
Component B	54.0g
Unpigmented Base	
Component C	85.0g
Reducer Base	

The above formula was used to prepare a silver metallic base coat. The order of addition is that listed. After each introduction of a component, the entire solution was stirred. The base coat was sprayed on panels primed with an epoxy primer. The base coat was sprayed on the substrate using conventional spray equipment until hiding was achieved. The base coat was allowed to flash for 30 minutes and then clear coated with a transparent coating based on isocyanate technology. The final properties of the base coat after coating with a conventual polyurethane clear coat are listed below, including appearance ratings.

## Properties of the Silver Metallic Base Coat and Base Coat, Clear Coat System

5	Property	Value
	Water	73.1 %
	Volatile Organic	7.7 %
10	Pigment	2.7 %
	Nonvolatile	19.2 %
	Binder	11.2 %
	Weight/Gallon (lbs/gal)	8.4
	VOC (lbs/gal)	2.6
15	Gloss, 20	89
	DOI	90
	Adhesion	good
	Film Thickness (mil)	0.53
	Humidity	good
20	Water Spray	excell.
	Metal Control	excell.

25 Example 4: Base Coat Preparation

The hydroxyl-functional acrylic polymer is prepared as follows. Approximately 695 grams of Toluene is combined with 175 grams of Urethane Grade Methyl Ethyl Ketone in a five-liter distillation flask, fitted with a water-cooled condenser, under a Nitrogen blanket. The mixture is brought to reflux, and adjusted to reflux at 100° C by adding Methyl Ethyl Ketone, the Nitrogen flow reduced to a minimum, and a 2215 grams of mixture of  $\beta$ -ethylenically unsaturated monomers is pumped into the flask over two hours. The monomer mixture consists of Methyl Methacrylate - 8.1%, Styrene - 21.2%, n-Butyl Methacrylate - 54.2%, Acrylic Acid - 2.7%, 2-Ethylhexyl Acrylate - 2.8%, and alpha-Methyl Styrene - 2.0%, and Hydroxyethyl Methacrylate - 9.1%. While the monomers are being pumped into the reaction flask, a mixture of 55 grams of free-radical initiator ( , 'azobis(isobutyronitrile)) and 250 grams solvent (Methyl Ethyl Ketone) is simultaneously fed into the flask over two hours.

After adding all the monomers, solvent, and initiator (and flushing the pumps and feed lines with propylene glycol monopropyl ether), the reaction is held at reflux. After sixty minutes at reflux, a mixture of 11 grams of initiator and 120 grams of solvent (Methyl Ethyl Ketone) is pumped into the reaction flask over seventy-five minutes. After adding the initiator and solvent mixture, the reaction is held at reflux. After one to two hours at reflux, the reaction vessel is cooled to just below reflux, and fitted with a Barrett trap below the condenser. Solvent is distilled off, until 130 grams of solvent is collected. Then the solution is cooled and poured into a lined metal can. The acid number of this polymer is between 19 and 21 milligrams of KOH per gram of resin solids, and its percent non-volatile is between 66% and 69%.

The diisocyanate / polyether adduct is formed by the reaction of one isocyanate group of a diisocyanate with the hydroxyl group of a Methoxy Polyethylene Glycol of the formula:  $\text{CH}_3\text{O}-(\text{CH}_2\text{CH}_2\text{O})_n\text{-H}$ , where  $n = 7 - 45$ . This reaction is carried out in a dry air atmosphere, as follows.

In a 1-liter distillation flask, fitted with a water-cooled condenser and a dry air-sparge tube, under a dry air atmosphere, 231.7 grams of isophorone diisocyanate is combined with 35.9 grams of Urethane Grade Methyl Ethyl Ketone, and 374.5 grams of Methoxy Polyethylene Glycol of the formula:  $\text{CH}_3\text{O}-(\text{CH}_2\text{CH}_2\text{O})_n\text{-H}$  (molecular weight = 350). The mixture is then stirred for 15 minutes at ambient temperature, and 0.2 grams of dibutyltin diacetate is added. The mixture is continuously stirred, and heated as necessary, to maintain a temperature of 60° C for two hours. The percent isocyanate of the adduct solution should be 6.7% - 6.9%, or 7.2% based on solids.

To make the final urethane / acrylic graft copolymer, 610.9 grams of the hydroxyl-functional acrylic polymer is combined with 94.6 grams of the diisocyanate / polyether adduct, and 60 grams of Toluene in a five-liter distillation flask, fitted with a water-cooled condenser, under a dry Nitrogen atmosphere. The mixture is heated to 75° C, and held at that temperature for two and one-half hours. The temperature is

then increased until the solution reaches reflux ( $\sim 104^{\circ}\text{C}$ ), and held at reflux for thirty minutes. The solution is then cooled to  $75^{\circ}\text{C}$ , and vacuum stripped (85 mbars pressure) to remove approximately 100 grams of solvent. The vacuum is broken with Nitrogen, and 220 grams of ethylene glycol monobutyl ether is added to the flask. After stirring the mixture for ten minutes, vacuum stripping is resumed. Stripping continues at  $75^{\circ}\text{C}$  / 85 mbars until a total of 210 to 212 grams (including the first 100 grams) of solvent has been removed. The solution is then cooled, and filtered into a lined can. The acid number is between 15 and 17 milligrams of KOH per gram of resin solids, and the percent non-volatile is between 66% and 69%.

#### Preparation of Component A - Aluminum Base

A commercial solvent borne aluminum paste, 60.0g, was added to 60.0g of butoxyethanol. The mixture was stirred until a homogeneous dispersion was produced. To this dispersion, 30.0g, of the resin described above was added to the dispersion and the entire mixture stirred until a smooth, uniform slurry was produced. This aluminum base contains 39% volatile organic solvent by weight, a solids level of 61.0% by weight, a binder level of 13% by weight and an aluminum flake level of 26.0% by weight.

#### Preparation of Component B - Pigmented Base

The following pigment slurry was prepared, 12.2g of NeoCryl A-6037, 14.8g of NeoCryl A-640, 0.3g of 30% aqueous ammonia, 3.2g of Perrindo Maroon R-6424 manufactured by Mobay Corporation, 0.1g of a defoamer and 0.05g of Surfynol 104BC manufactured by Air Products. The above mixture were mixed together and predispersed using a Cowles blade. The mixture was then ground on a sand mill until the desired particle size was achieved. A let down solution containing 29.5g of NeoCryl A-622, 2.6g of butoxyethanol, 1.3g of deionized water and 0.03g of 1% aqueous solution of FC-120 was added to the grind portion under agitation. An unpigmented base can be prepared using the procedure above with the elimination of the Perrindo Maroon R-6424 pigment. A wax dispersion was prepared by grinding 50.0g of a wax, 50.0g of unpigmented resin solution, 50.0 g of deionized water and 50.0g of butoxyethanol. This paste was ground to the desired particle size using conventional techniques. To the pigmented and unpigmented bases described above, 6.0g of this wax dispersion was added with stirring. Also added to the pigmented and unpigmented bases described above was 16.0g of NeoRez R-9637 manufactured by ICI Resins. The resulting pigmented base had a pH of 8.4, a solids level of 38.1% by weight, a volatile organic level of 10.5% by weight, a binder level of 34.2% by weight, a pigment level of 4.3% by weight and a water level of 51.4% by weight. The unpigmented base has a solid level of 35.4% by weight, volatile organic level of 9.5% by weight, a water level of 55.1% by weight and a pH of 8.5.

#### Preparation of Component C - Reducer Base

To 1953.2g of deionized water, 31.3g of Laponite RD manufactured by Laporte Industries and 10.7g of Pluracol P1010 manufactured by BASF Corporation was added. The resulting solution was then agitated for one hour until the Laponite RD was completely dispersed. To this solution was added 1953.2g of deionized water. The resulting mixture had a Laponite RD level of 0.79% by weight and a polypropylene glycol level of 0.27% by weight.

#### EXAMPLE 5 - PREPARATION OF A WATER BORNE RED METALLIC BASE COAT

Component	Parts by Weight
Component A Aluminum Base	16.0
Component B Pigmented Base	54.0
Component C Reducer Base	85.0

The above formula was used to prepare a red metallic base coat. The order of addition is that listed. After each introduction of a component, the entire solution was stirred. The base coat was sprayed on panels primed with an epoxy primer. The base coat was sprayed on the substrate using conventional spray equipment until hiding was achieved. The base coat was allowed to flash for 30 minutes and then clear coated with a transparent coating based on isocyanate technology. The final properties of the base coat after coating with a conventual polyurethane clear coat are listed below, including appearance ratings.

#### Properties of the Red Metallic Base Coat and Base Coat, Clear Coat System

Property	Value
Water	72.5 %
Volatile Organic	7.4 %
Pigment	2.4 %
Nonvolatile	20.1 %
Binder	10.8 %
Weight per Gallon (lbs/gal)	8.4
VOC (lbs/gal)	2.4
Gloss, 20 deg	90
DOI	84
Adhesion	good
Film Thickness (mil)	0.34
Humidity	good
Water Spray	excell.
Metal Control	excell.

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Component A 16.0g  
Aluminum Base

Component B 54.0g  
Unpigmented Base

Component C 85.0g  
Reducer Base

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The above formula was used to prepare a silver metallic base coat. The order of addition is that listed. After each introduction of a component, the entire solution was stirred. The base coat was sprayed on panels primed with an epoxy primer. The base coat was sprayed on the substrate using conventional spray equipment until hiding was achieved. The base coat was allowed to flash for 30 minutes and then clear coated with a transparent coating based on isocyanate technology. The final properties of the base coat after coating with a conventional polyurethane clear coat are listed below, including appearance ratings.

Properties of the Silver Metallic Base Coat and Base Coat, Clear Coat System

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Property	Value
Water	73.1 %
Volatile Organic	7.7 %
Pigment	2.7 %
Nonvolatile	19.2 %
Binder	11.2 %
Weight/Gallon (lbs/gal)	8.4
VOC (lbs/gal)	2.6
Gloss, 20	89
DOI	90
Adhesion	good
Film Thickness (mil)	0.53
Humidity	good
Water Spray	excell.
Metal Control	excell.

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Example 7. Base Coat Preparation

Approximately 134 grams of Propylene glycol monopropyl ether is combined with 5 grams of Methyl Ethyl Ketone in a one-liter distillation flask, fitted with a water-cooled condenser, under a Nitrogen blanket. The mixture is brought to reflux ( $\sim 140^{\circ}\text{C}$ ), the Nitrogen flow reduced to a minimum, and 454 grams of a mixture of  $\alpha,\beta$ -ethylenically unsaturated monomers is pumped into the flask over two hours. The monomer mixture consists of Methyl Methacrylate - 7.2%, Styrene - 18.8%, n-Butyl Methacrylate - 48.1%, Acrylic Acid - 1.7%, 2-Ethylhexyl Acrylate - 2.4%, alpha-Methyl Styrene - 1.7%, and Polyethylene Glycol Monomethacrylate (containing 10 Ethylene oxide units) - 19.9%. While the monomers are being pumped into the reaction flask, a mixture of 11 grams of free-radical initiator (t-Butyl Perbenzoate) and 50 grams solvent (propylene glycol monopropyl ether) is simultaneously fed into the flask over two hours.

After adding all the monomers, solvent, and initiator (and flushing the pumps and feed lines with propylene glycol monopropyl ether), the reaction is held at reflux. After thirty minutes at reflux, a mixture of 1.7 grams of initiator and 35 grams of solvent (propylene glycol monopropyl ether) is pumped into the reaction flask over ten minutes. After adding the initiator and solvent mixture, the reaction is held at reflux. After one to two hours at reflux, the reaction is cooled and filtered into a lined metal can. The viscosity of this polymer is between 6,000 and 8,000 cps at 63.4% non-volatile by weight, measured at  $25^{\circ}\text{C}$ , and its acid number is from 14 to 16 milligrams of KOH per gram of resin solids.

#### Preparation of Component A - Aluminum Base

A commercial solvent borne aluminum paste, 60.0g, was added to 60.0g of butoxyethanol. The mixture was stirred until a homogeneous dispersion was produced. To this dispersion, 30.0g, of the resin described above was added to the dispersion and the entire mixture stirred until a smooth, uniform slurry was produced. This aluminum base contains 39% volatile organic solvent by weight, a solids level of 61.0% by weight, a binder level of 13% by weight and an aluminum flake level of 26.0% by weight.

#### Preparation of Component B - Pigmented Base

The following pigment slurry was prepared, 12.2g of NeoCryl A-6037, 14.8g of NeoCryl A-640, 0.3g of 30% aqueous ammonia, 3.2g of Perrindo Maroon R-6424 manufactured by Mobay Corporation, 0.1g of a defoamer and 0.05g of Surfynol 104BC manufactured by Air Products. The above mixture were mixed together and predispersed using a Cowles blade. The mixture was then ground on a sand mill until the desired particle size was achieved. A let down solution containing 29.5g of NeoCryl A-622, 2.6g of butoxyethanol, 1.3g of deionized water and 0.03g of 1% aqueous solution of FC-120 was added to the grind portion under agitation. An unpigmented base can be prepared using the procedure above with the elimination of the Perrindo Maroon R-6424 pigment. A wax dispersion was prepared by grinding 50.0g of a wax, 50.0g of unpigmented resin solution, 50.0 g of deionized water and 50.0g of butoxyethanol. This paste was ground to the desired particle size using conventional techniques. To the pigmented and unpigmented bases described above, 6.0g of this wax dispersion was added with stirring. The resulting pigmented base had a pH of 8.6, a solids level of 37.8% by weight, a volatile organic level of 10.6% by weight, a binder level of 33.3% by weight, a pigment level of 5.0% by weight and a water level of 51.6% by weight. The unpigmented base has a solid level of 34.7% by weight, volatile organic level of 9.4% by weight, a water level of 55.9% by weight and a pH of 8.7.

#### Preparation of Component C - Reducer Base

To 1953.2g of deionized water, 31.3g of Laponite RD manufactured by Laporte Industries was added. The resulting solution was then agitated for one hour until the Laponite RD was completely dispersed. To this solution was added 8.0g of Troysol LAC manufactured by Troy Chemical Company and the entire solution stirred for 15 minutes. The resulting solution was then diluted with 1953.2g of deionized water. To this Laponite RD dispersion was added 462.9g of NeoRex R-9637 manufactured by ICI Resins. The resulting mixture had a Laponite RD level of 0.71% by weight, Troysol LAC level of 0.18% by weight, water dispersible polyurethane level of 4.2% by weight and a volatile organic level of 1.1% by weight.

#### EXAMPLE 8 - PREPARATION OF A WATER BORNE RED METALLIC BASE COAT

Component	Parts by Weight
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Component A Aluminum Base	16.0
Component B Pigmented Base	44.0
Component C Reducer Base	95.0

The above formula was used to prepare a red metallic base coat. The order of addition is that listed. After each introduction of a component, the entire solution was stirred. The base coat was sprayed on panels primed with an epoxy primer. The base coat was sprayed on the substrate using conventional spray equipment until hiding was achieved. The base coat was allowed to flash for 30 minutes and then clear coated with a transparent coating based on isocyanate technology. The final properties of the base coat after coating with a conventual polyurethane clear coat are listed below, including appearance ratings.

## Properties of the Red Metallic Base Coat and Base Coat, Clear Coat System

Property	Value
Water	72.5 %
Volatile Organic	7.4 %
Pigment	2.4 %
Nonvolatile	20.1 %
Binder	10.8 %
Weight per Gallon (lbs/gal)	8.4
VOC (lbs/gal)	2.2
Gloss, 20 deg	92
DOI	87
Adhesion	good
Film Thickness (mil)	0.29
Humidity	good
Water Spray	excell.
Metal Control	excell.

## EXAMPLE 9 - Preparation of a Water Borne Silver Metallic Base Coat

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5	Component A	16.0g
	Aluminum Base	
	Component B	44.0g
	Unpigmented Base	
10	Component C	95.0g
	Reducer Base	

---

15 The above formula was used to prepare a silver metallic base coat. The order of addition is that listed. After each introduction of a component, the entire solution was stirred. The base coat was sprayed on panels primed with an epoxy primer. The base coat was sprayed on the substrate using conventional spray equipment until hiding was achieved. The base coat was allowed to flash for 30 minutes and then clear coated with a transparent coating based on isocyanate technology. The final properties of the base coat after coating with a conventional polyurethane clear coat are listed below, including appearance ratings.

25 Properties of the Silver Metallic Base Coat and Base Coat. Clear Coat System

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Property	Value
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30	Water	73.1 %
	Volatile Organic	7.7 %
	Pigment	2.7 %
35	Nonvolatile	19.2 %
	Binder	11.2 %
	Weight/Gallon (lbs/gal)	8.45
	VOC (lbs/gal)	2.4
40	Gloss, 20	93

45	DOI	86
	Adhesion	good
	Film Thickness (mil)	0.37
	Humidity	good
	Water Spray	excell.
50	Metal Control	excell.

Example 10: Base Coat Preparation

55 Approximately 134 grams of Propylene glycol monopropyl ether is combined with 5 grams of Methyl Ethyl Ketone in a one-liter distillation flask, fitted with a water-cooled condenser, under a Nitrogen blanket. The mixture is brought to reflux (~ 140° C), the Nitrogen flow reduced to a minimum, and 454 grams of a



mixture of  $\beta$ -ethylenically unsaturated monomers is pumped into the flask over two hours. The monomer mixture consists of Methyl Methacrylate - 7.2%, Styrene - 18.8%, n-Butyl Methacrylate - 48.1%, Acrylic Acid - 1.7%, 2-Ethylhexyl Acrylate - 2.4%, alpha-Methyl Styrene - 1.7%, and Polyethylene Glycol Monomethacrylate (containing 10 Ethylene oxide units) - 19.9%. While the monomers are being pumped into the reaction flask, a mixture of 11 grams of free-radical initiator (t-Butyl Perbenzoate) and 50 grams solvent (propylene glycol monopropyl ether) is simultaneously fed into the flask over two hours.

After adding all the monomers, solvent, and initiator (and flushing the pumps and feed lines with propylene glycol monopropyl ether), the reaction is held at reflux. After thirty minutes at reflux, a mixture of 1.7 grams of initiator and 35 grams of solvent (propylene glycol monopropyl ether) is pumped into the reaction flask over ten minutes. After adding the initiator and solvent mixture, the reaction is held at reflux. After one to two hours at reflux, the reaction is cooled and filtered into a lined metal can. The viscosity of this polymer is between 6,000 and 8,000 cps at 63.4% non-volatile by weight, measured at 25 °C, and its acid number is from 14 to 16 milligrams of KOH per gram of resin solids.

#### Preparation of Component A - Aluminum Base

A commercial solvent borne aluminum paste, 60.0g, was added to 60.0g of butoxyethanol. The mixture was stirred until a homogeneous dispersion was produced. To this dispersion, 30.0g of the resin described above was added to the dispersion and the entire mixture stirred until a smooth, uniform slurry was produced. This aluminum base contains 39% volatile organic solvent by weight, a solids level of 61.0% by weight, a binder level of 13% by weight and an aluminum flake level of 26.0% by weight.

#### Preparation of Component B - Pigmented Base

The following pigment slurry was prepared, 12.2g of NeoCryl A-6037, 14.8g of NeoCryl A-640, 0.3g of 30% aqueous ammonia, 3.2g of Perrindo Maroon R-6424 manufactured by Mobay Corporation, 0.1g of a defoamer and 0.05g of Surfynol 104BC manufactured by Air Products. The above mixture were mixed together and predispersed using a Cowles blade. The mixture was then ground on a sand mill until the desired particle size was achieved. A let down solution containing 29.5g of NeoCryl A-622, 2.6g of butoxyethanol, 1.3g of deionized water and 0.03g of 1% aqueous solution of FC-120 was added to the grind portion under agitation. An unpigmented base can be prepared using the procedure above with the elimination of the Perrindo Maroon R-6424 pigment. A wax dispersion was prepared by grinding 50.0g of a wax, 50.0g of unpigmented resin solution, 50.0 g of deionized water and 50.0g of butoxyethanol. This paste was ground to the desired particle size using conventional techniques. To the pigmented and unpigmented bases described above, 6.0g of this wax dispersion was added with stirring. Also added to the pigmented and unpigmented bases described above was 16.0g of NeoRez R-9637 manufactured by ICI Resins. The resulting pigmented base had a pH of 8.4, a solids level of 38.1% by weight, a volatile organic level of 10.5% by weight, a binder level of 34.2% by weight, a pigment level of 4.3% by weight and a water level of 51.4% by weight. The unpigmented base has a solid level of 35.4% by weight, volatile organic level of 9.5% by weight, a water level of 55.1% by weight and a pH of 8.5.

#### Preparation of Component C - Reducer Base

To 1953.2g of deionized water, 31.3g of Laponite RD manufactured by Laporte Industries and 10.7g of Pluracol P1010 manufactured by BASF Corporation was added. The resulting solution was then agitated for one hour until the Laponite RD was completely dispersed. To this solution was added 1953.2g of deionized water. The resulting mixture had a Laponite RD level of 0.79% by weight and a polypropylene glycol level of 0.27% by weight.

#### EXAMPLE 11 - PREPARATION OF A WATER BORNE RED METALLIC BASE COAT

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Component	Parts by Weight
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Component A	16.0
Aluminum Base	
Component B	54.0
Pigmented Base	
Component C	85.0
Reducer Base	

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The above formula was used to prepare a red metallic base coat. The order of addition is that listed. After each introduction of a component, the entire solution was stirred. The base coat was sprayed on panels primed with an epoxy primer. The base coat was sprayed on the substrate using conventional spray equipment until hiding was achieved. The base coat was allowed to flash for 30 minutes and then clear coated with a transparent coating based on isocyanate technology. The final properties of the base coat after coating with a conventual polyurethane clear coat are listed below, including appearance ratings.

Properties of the Red Metallic Base Coat and Base Coat, Clear Coat System

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Property	Value
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Water	72.5 %
Volatile Organic	7.4 %
Pigment	2.4 %
Nonvolatile	20.1 %
Binder	10.8 %
Weight per Gallon (lbs/gal)	8.4
VOC (lbs/gal)	2.4
Gloss, 20 deg	90
DOI	84

Adhesion	good
Film Thickness (mil)	0.34
Humidity	good
Water Spray	excell.
Metal Control	excell.

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EXAMPLE 12 - Preparation of a Water Borne Silver Metallic Base Coat


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Component A	16.0g
Aluminum Base	
Component B	54.0g
Unpigmented Base	
Component C	85.0g
Reducer Base	

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The above formula was used to prepare a silver metallic base coat. The order of addition is that listed. After each introduction of a component, the entire solution was stirred. The base coat was sprayed on panels primed with an epoxy primer. The base coat was sprayed on the substrate using conventional spray equipment until hiding was achieved. The base coat was allowed to flash for 30 minutes and then clear coated with a transparent coating based on isocyanate technology. The final properties of the base coat after coating with a conventional polyurethane clear coat are listed below, including appearance ratings.

## Properties of the Silver Metallic Base Coat and Base Coat, Clear Coat System

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Property	Value
Water	73.1 %
Volatile Organic	7.7 %
Pigment	2.7 %
Nonvolatile	19.2 %
Binder	11.2 %
Weight/Gallon (lbs/gal)	8.4
VOC (lbs/gal)	2.6
Gloss, 20	89
DOI	90
Adhesion	good
Film Thickness (mil)	0.53
Humidity	good
Water Spray	excell.
Metal Control	excell.

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## Example 13: Base Coat Preparation

In a 1-liter distillation flask, fitted with a water-cooled condenser and a dry air-spargue tube, under a dry air atmosphere, 177.8 grams of isophorone diisocyanate is combined with 422.4 grams of Methoxy Polyethylene Glycol of the formula:  $\text{CH}_3\text{O}-(\text{CH}_2\text{CH}_2\text{O})_n\text{H}$  (molecular weight = 550), 10 grams of Urethane Grade Methyl Ethyl Ketone, and 12 grams of a 0.5% solution of Butylated Hydroxy Toluene in Urethane

Grade Methyl Ethyl Ketone.

The mixture is then stirred for 5 minutes at ambient temperature, and 0.2 grams of dibutyltin diacetate is added. The mixture is continuously stirred, and heated as necessary, to maintain a temperature of 60° C for two hours. The mixture is then cooled to less than 30° C, and 96.5 grams of Hydroxyethyl Acrylate is added. The mixture is continuously stirred, and heated as necessary, to maintain a temperature of 60° C for two hours. Any residual isocyanate (as determined by Infrared Spectroscopy) should be reacted with a minimum amount of anhydrous ethanol. The product can be sealed in a dried glass bottle.

Approximately 134 grams of Propylene glycol monopropyl ether is combined with 5 grams of Methyl Ethyl Ketone in a one-liter distillation flask, fitted with a water-cooled condenser, under a Nitrogen blanket. The mixture is brought to reflux (~ 140° C), the Nitrogen flow reduced to a minimum, and a mixture of  $\beta$ -ethylenically unsaturated monomers is pumped into the flask over two hours. The monomer mixture consists of Methyl Methacrylate - 7.3%, Styrene - 18.9%, n-Butyl Methacrylate - 47.9%, Acrylic Acid - 1.6%, 2-Ethylhexyl Acrylate - 2.4%, and Polyether Monomer (described above) - 22.0%. While the monomers are being pumped into the reaction flask, a mixture of 10.6 grams of free-radical initiator (t-Butyl Perbenzoate) and 50 grams solvent (propylene glycol monopropyl ether) is simultaneously fed into the flask over two hours.

After adding all the monomers, solvent, and initiator (and flushing the pumps and feed lines with propylene glycol monopropyl ether), the reaction is held at reflux. After thirty minutes at reflux, a mixture of 1.2 grams of initiator and 35 grams of solvent (propylene glycol monopropyl ether) is pumped into the reaction flask over ten minutes. After adding the initiator and solvent mixture, the reaction is held at reflux. After one to two hours at reflux, the reaction is cooled and filtered into a lined metal can. The viscosity of this polymer is between 2,500 and 4,500 cps at 66.2% non-volatile by weight, measured at 25° C, and its acid number is from 12.5 to 14.5 milligrams of KOH per gram of resin solids.

#### Preparation of Component A - Aluminum Base

A commercial solvent borne aluminum paste, 60.0g, was added to 60.0g of butoxyethanol. The mixture was stirred until a homogeneous dispersion was produced. To this dispersion, 30.0g, of the resin described above was added to the dispersion and the entire mixture stirred until a smooth, uniform slurry was produced. This aluminum base contains 39% volatile organic solvent by weight, a solids level of 61.0% by weight, a binder level of 13% by weight and an aluminum flake level of 26.0% by weight.

#### Preparation of Component B - Pigmented Base

The following pigment slurry was prepared, 12.2g of NeoCryl A-6037, 14.8g of NeoCryl A-640, 0.3g of 30% aqueous ammonia, 3.2g of Perrindo Maroon R-6424 manufactured by Mobay Corporation, 0.1g of a defoamer and 0.05g of Surfynol 104BC manufactured by Air Products. The above mixture were mixed together and predispersed using a Cowles blade. The mixture was then ground on a sand mill until the desired particle size was achieved. A let down solution containing 29.5g of NeoCryl A-622, 2.6g of butoxyethanol, 1.3g of deionized water and 0.03g of 1% aqueous solution of FC-120 was added to the grind portion under agitation. An unpigmented base can be prepared using the procedure above with the elimination of the Perrindo Maroon R-6424 pigment. A wax dispersion was prepared by grinding 50.0g of a wax, 50.0g of unpigmented resin solution, 50.0 g of deionized water and 50.0g of butoxyethanol. This paste was ground to the desired particle size using conventional techniques. To the pigmented and unpigmented bases described above, 6.0g of this wax dispersion was added with stirring. The resulting pigmented base had a pH of 8.6, a solids level of 37.8% by weight, a volatile organic level of 10.6% by weight, a binder level of 33.3% by weight, a pigment level of 5.0% by weight and a water level of 51.6% by weight. The unpigmented base has a solid level of 34.7% by weight, volatile organic level of 9.4% by weight, a water level of 55.9% by weight and a pH of 8.7.

#### Preparation of Component C - Reducer Base

To 1953.2g of deionized water, 31.3g of Laponite RD manufactured by Laporte Industries was added. The resulting solution was then agitated for one hour until the Laponite RD was completely dispersed. To this solution was added 8.0g of Troysol LAC manufactured by Troy Chemical Company and the entire solution stirred for 15 minutes. The resulting solution was then diluted with 1953.2g of deionized water. To this Laponite RD dispersion was added 462.9g of NeoRez R-9637 manufactured by ICI Resins. The resulting mixture had a Laponite RD level of 0.71% by weight, Troysol LAC level of 0.18% by weight, water

dispersible polyurethane level of 4.2% by weight and a volatile organic level of 1.1% by weight.

#### EXAMPLE 14 - PREPARATION OF A WATER BORNE RED METALLIC BASE COAT

Component	Parts by Weight
Component A Aluminum Base	16.0
Component B Pigmented Base	44.0
Component C Reducer Base	95.0

The above formula was used to prepare a red metallic base coat. The order of addition is that listed. After each introduction of a component, the entire solution was stirred. The base coat was sprayed on panels primed with an epoxy primer. The base coat was sprayed on the substrate using conventional spray equipment until hiding was achieved. The base coat was allowed to flash for 30 minutes and then clear coated with a transparent coating based on isocyanate technology. The final properties of the base coat after coating with a conventual polyurethane clear coat are listed below, including appearance ratings.

#### Properties of the Red Metallic Base Coat and Base Coat, Clear Coat System

Property	Value
Water	72.5 %
Volatile Organic	7.4 %
Pigment	2.4 %
Nonvolatile	20.1 %
Binder	10.8 %
Weight per Gallon (lbs/gal)	8.4
VOC (lbs/gal)	2.2
Gloss, 20 deg	92
DOI	87
Adhesion	good
Film Thickness (mil)	0.29
Humidity	good
Water Spray	excell.
Metal Control	excell.

EXAMPLE 15 - Preparation of a Water Borne Silver Metallic Base Coat


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5	Component A	16.0g
	Aluminum Base	
10	Component B	44.0g
	Unpigmented Base	
	Component C	95.0g
15	Reducer Base	

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20 The above formula was used to prepare a silver metallic base coat. The order of addition is that listed. After each introduction of a component, the entire solution was stirred. The base coat was sprayed on panels primed with an epoxy primer. The base coat was sprayed on the substrate using conventional spray equipment until hiding was achieved. The base coat was allowed to flash for 30 minutes and then clear coated with a transparent coating based on isocyanate technology. The final properties of the base coat

25 after coating with a conventional polyurethane clear coat are listed below, including appearance ratings.

## Properties of the Silver Metallic Base Coat and Base Coat, Clear Coat System

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30	Property	Value
	Water	73.1 %
35	Volatile Organic	7.7 %
	Pigment	2.7 %
	Nonvolatile	19.2 %
	Binder	11.2 %
	Weight/Gallon (lbs/gal)	8.45
40	VOC (lbs/gal)	2.4
	Gloss, 20	93
	DOI	86
	Adhesion	good
	Film Thickness (mil)	0.37
45	Humidity	good
	Water Spray	excell.
	Metal Control	excell.

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Example 16: Base Coat Preparation

55 In a 1-liter distillation flask, fitted with a water-cooled condenser and a dry air-sparge tube, under a dry air atmosphere, 178 grams of meta-isopropenyl- $\alpha,\alpha$ -dimethylbenzyl isocyanate is combined with 484.8 grams of Methoxy Polyethylene Glycol of the formula:  $\text{CH}_3\text{O}-(\text{CH}_2\text{CH}_2\text{O})_n\text{H}$  (molecular weight = 550), 60 grams of Urethane Grade Methyl Ethyl Ketone, and 13.2 grams of a 0.5% solution of Butylated Hydroxy

Toluene in Urethane Grade Methyl Ethyl Ketone. The mixture is then stirred for 5 minutes at ambient temperature, and 0.2 grams of dibutyltin diacetate is added. The mixture is continuously stirred, and heated as necessary, to maintain a temperature of 60 ° C for two hours. Any residual isocyanate (as determined by Infrared Spectroscopy) should be reacted with a minimum amount of anhydrous ethanol. The product can be sealed in a dried glass bottle.

Approximately 134 grams of Propylene glycol monopropyl ether is combined with 5 grams of Methyl Ethyl Ketone in a one-liter distillation flask, fitted with a water-cooled condenser, under a Nitrogen blanket. The mixture is brought to reflux (~ 140 ° C), the Nitrogen flow reduced to a minimum, and a mixture of  $\alpha,\beta$ -ethylenically unsaturated monomers is pumped into the flask over two hours. The monomer mixture consists of Methyl Methacrylate - 7.3%, Styrene - 18.9%, n-Butyl Methacrylate - 47.9%, Acrylic Acid - 1.6%, 2-Ethylhexyl Acrylate - 2.4%, and Polyether Monomer (described above) - 22.0%. While the monomers are being pumped into the reaction flask, a mixture of 10.6 grams of free-radical initiator (t-Butyl Perbenzoate) and 50 grams solvent (propylene glycol monopropyl ether) is simultaneously fed into the flask over two hours.

After adding all the monomers, solvent, and initiator (and flushing the pumps and feed lines with propylene glycol monopropyl ether), the reaction is held at reflux. After thirty minutes at reflux, a mixture of 1.2 grams of initiator and 35 grams of solvent (propylene glycol monopropyl ether) is pumped into the reaction flask over ten minutes. After adding the initiator and solvent mixture, the reaction is held at reflux. After one to two hours at reflux, the reaction is cooled and filtered into a lined metal can. The viscosity of this polymer is between 2,500 and 4,500 cps at 66.2% non-volatile by weight, measured at 25 ° C, and its acid number is from 12.5 to 14.5 milligrams of KOH per gram of resin solids.

#### Preparation of Component A - Aluminum Base

A commercial solvent borne aluminum paste, 60.0g, was added to 60.0g of butoxyethanol. The mixture was stirred until a homogeneous dispersion was produced. To this dispersion, 30.0g. of the resin described above was added to the dispersion and the entire mixture stirred until a smooth, uniform slurry was produced. This aluminum base contains 39% volatile organic solvent by weight, a solids level of 61.0% by weight, a binder level of 13% by weight and an aluminum flake level of 26.0% by weight.

#### Preparation of Component B - Pigmented Base

The following pigment slurry was prepared, 12.2g of NeoCryl A-6037, 14.8g of NeoCryl A-640, 0.3g of 30% aqueous ammonia, 3.2g of Perrindo Maroon R-6424 manufactured by Mobay Corporation, 0.1g of a defoamer and 0.05g of Surfynol 104BC manufactured by Air Products. The above mixture were mixed together and predispersed using a Cowles blade. The mixture was then ground on a sand mill until the desired particle size was achieved. A let down solution containing 29.5g of NeoCryl A-622, 2.6g of butoxyethanol, 1.3g of deionized water and 0.03g of 1% aqueous solution of FC-120 was added to the grind portion under agitation. An unpigmented base can be prepared using the procedure above with the elimination of the Perrindo Maroon R-6424 pigment. A wax dispersion was prepared by grinding 50.0g of a wax, 50.0g of unpigmented resin solution, 50.0 g of deionized water and 50.0g of butoxyethanol. This paste was ground to the desired particle size using conventional techniques. To the pigmented and unpigmented bases described above, 6.0g of this wax dispersion was added with stirring. The resulting pigmented base had a pH of 8.6, a solids level of 37.8% by weight, a volatile organic level of 10.6% by weight, a binder level of 33.3% by weight, a pigment level of 5.0% by weight and a water level of 51.6% by weight. The unpigmented base has a solid level of 34.7% by weight, volatile organic level of 9.4% by weight, a water level of 55.9% by weight and a pH of 8.7.

#### Preparation of Component C - Reducer Base

To 1953.2g of deionized water, 31.3g of Laponite RD manufactured by Laporte Industries was added. The resulting solution was then agitated for one hour until the Laponite RD was completely dispersed. To this solution was added 8.0g of Troysol LAC manufactured by Troy Chemical Company and the entire solution stirred for 15 minutes. The resulting solution was then diluted with 1953.2g of deionized water. To this Laponite RD dispersion was added 462.9g of NeoRez R-9637 manufactured by ICI Resins. The resulting mixture had a Laponite RD level of 0.71% by weight, Troysol LAC level of 0.18% by weight, water dispersible polyurethane level of 4.2% by weight and a volatile organic level of 1.1% by weight.

EP 0 471 972 A2

EXAMPLE 17 - PREPARATION OF A WATER BORNE RED METALLIC BASE COAT

Component	Parts by Weight
Component A Aluminum Base	16.0
Component B Pigmented Base	44.0
Component C Reducer Base	95.0

The above formula was used to prepare a red metallic base coat. The order of addition is that listed. After each introduction of a component, the entire solution was stirred. The base coat was sprayed on panels primed with an epoxy primer. The base coat was sprayed on the substrate using conventional spray equipment until hiding was achieved. The base coat was allowed to flash for 30 minutes and then clear coated with a transparent coating based on isocyanate technology. The final properties of the base coat after coating with a conventual polyurethane clear coat are listed below, including appearance ratings.

Properties of the Red Metallic Base Coat and Base Coat, Clear Coat System

Property	Value
Water	72.5 %
Volatile Organic	7.4 %
Pigment	2.4 %
Nonvolatile	20.1 %
Binder	10.8 %
Weight per Gallon (lbs/gal)	8.4
VOC (lbs/gal)	2.2
Gloss, 20 deg	92
DOI	87



Adhesion	good
Film Thickness (mil)	0.29
Humidity	good
Water Spray	excell.
Metal Control	excell.

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EXAMPLE 18 - Preparation of a Water Borne Silver Metallic Base Coat

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Component A	16.0g
Aluminum Base	
Component B	44.0g
Unpigmented Base	
Component C	95.0g
Reducer Base	

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The above formula was used to prepare a silver metallic base coat. The order of addition is that listed. After each introduction of a component, the entire solution was stirred. The base coat was sprayed on panels primed with an epoxy primer. The base coat was sprayed on the substrate using conventional spray equipment until hiding was achieved. The base coat was allowed to flash for 30 minutes and then clear coated with a transparent coating based on isocyanate technology. The final properties of the base coat after coating with a conventional polyurethane clear coat are listed below, including appearance ratings.

Properties of the Silver Metallic Base Coat and Base Coat, Clear Coat System

Property	Value
Water	73.1 %
Volatile Organic	7.7 %
Pigment	2.7 %
Nonvolatile	19.2 %
Binder	11.2 %
Weight/Gallon (lbs/gal)	8.45
VOC (lbs/gal)	2.4
Gloss, 20	93
DOI	86
Adhesion	good
Film Thickness (mil)	0.37
Humidity	good
Water Spray	excell.
Metal Control	excell.

cured with polyisocyanates. Other examples of cured transparent clear coats are acrylic and/or polyester resins cured with silanes, acrylic and/or polyester acid cured epoxy coatings, and acrylic and/or polyester resins cured with aminoplast resins, however, this is not an inclusive list.

#### PREFERRED EMBODIMENTS OF INVENTION

The following examples illustrate the invention without limiting the scope thereof.

##### Example 19 Base Coat Preparation

Approximately 134 grams of Propylene glycol monopropyl ether is combined with 5 grams of Methyl Ethyl Ketone in a one-liter distillation flask, fitted with a water-cooled condenser, under a Nitrogen blanket. The mixture is brought to reflux (~ 140° C), the Nitrogen flow reduced to a minimum, and 454 grams of a mixture of  $\beta$ -ethylenically unsaturated monomers is pumped into the flask over two hours. The monomer mixture consists of Methyl Methacrylate - 7.2%, Styrene - 18.8%, n-Butyl Methacrylate - 48.1%, Acrylic Acid - 1.7%, 2-Ethylhexyl Acrylate - 2.4%, alpha-Methyl Styrene - 1.7%, and Polyethylene Glycol Monomethacrylate (containing 10 Ethylene oxide units) - 19.9%.

While the monomers are being pumped into the reaction flask, a mixture of 11 grams of free-radical initiator (t-Butyl Perbenzoate) and 50 grams solvent (propylene glycol monopropyl ether) is simultaneously fed into the flask over two hours.

After adding all the monomers, solvent, and initiator (and flushing the pumps and feed lines with propylene glycol monopropyl ether), the reaction is held at reflux. After thirty minutes at reflux, a mixture of 1.7 grams of initiator and 35 grams of solvent (propylene glycol monopropyl ether) is pumped into the reaction flask over ten minutes. After adding the initiator and solvent mixture, the reaction is held at reflux. After one to two hours at reflux, the reaction is cooled and filtered into a lined metal can. The viscosity of this polymer is between 6,000 and 8,000 cps at 63.4% non-volatile by weight, measured at 25° C, and its acid number is from 14 to 16 milligrams of KOH per gram of resin solids.

##### Preparation of Component A - Aluminum Base

A commercial solvent borne aluminum paste, 60.0g, was added to 60.0g of butoxyethanol. The mixture was stirred until a homogeneous dispersion was produced. To this dispersion, 30.0g, of the resin described above was added to the dispersion and the entire mixture stirred until a smooth, uniform slurry was produced. This aluminum base contains 39% volatile organic solvent by weight, a solids level of 61.0% by weight, a binder level of 13% by weight and an aluminum flake level of 26.0% by weight.

Preparation of Component B - Pigmented Base

The following pigment slurry was prepared, 12.2g of NeoCryl A-6037, 14.8g of NeoCryl A-640, 0.3g of 30% aqueous ammonia, 3.2g of Perrindo Maroon R-6424 manufactured by Mobay Corporation, 0.1g of a defoamer and 0.05g of Surfynol 104BC manufactured by Air Products. The above mixture were mixed together and predispersed using a Cowles blade. The mixture was then ground on a sand mill until the desired particle size was achieved. A let down solution containing 29.5g of NeoCryl A-622, 2.6g of butoxyethanol, 1.3g of deionized water and 0.03g of 1% aqueous solution of FC-120 was added to the grind portion under agitation. An unpigmented base can be prepared using the procedure above with the elimination of the Perrindo Maroon R-6424 pigment. A wax dispersion was prepared by grinding 50.0g of a wax, 50.0g of unpigmented resin solution, 50.0 g of deionized water and 50.0g of butoxyethanol. This paste was ground to the desired particle size using conventional techniques. To the pigmented and unpigmented bases described above, 6.0g of this wax dispersion was added with stirring. Also added to the pigmented and unpigmented bases described above was 16.0g of NooRez R-9637 manufactured by ICI Resins. The resulting pigmented base had a pH of 8.4, a solids level of 38.1% by weight, a volatile organic level of 10.5% by weight, a binder level of 34.2% by weight, a pigment level of 4.3% by weight and a water level of 51.4% by weight. The unpigmented base has a solid level of 35.4% by weight, volatile organic level of 9.5% by weight, a water level of 55 1% by weight and a pH of 8.5.

Preparation of Component C - Reducer Base

To 1953.2g of deionized water, 31.3g of Laponite RD manufactured by Laporte Industries and 10.7g of Pluracol P1010 manufactured by BASF Corporation was added. The resulting solution was then agitated for one hour until the Laponite RD was completely dispersed. To this solution was added 1953.2g of deionized water. The resulting mixture had a Laponite RD level of 0.79% by weight and a polypropylene glycol level of 0.27% by weight.

EXAMPLE 20 - PREPARATION OF A WATER BORNE RED METALLIC BASE COAT

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Component	Parts by Weight
Component A Aluminum Base	16.0
Component B Pigmented Base	54.0
Component C Reducer Base	85.0

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The above formula was used to prepare a red metallic base coat. The order of addition is that listed. After each introduction of a component, the entire solution was stirred. The base coat was sprayed on panels primed with an epoxy primer. The base coat was sprayed on the substrate using conventional spray equipment until hiding was achieved. The base coat was allowed to flash for 30 minutes and then clear coated with a transparent coating based on isocyanate technology. The final properties of the base coat after coating with a conventual polyurethane clear coat are listed below, including appearance ratings.

Properties of the Red Metallic Base Coat and Base Coat, Clear Coat System

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Property	Value
Water	72.5 %
Volatile Organic	7.4 %
Pigment	2.4 %
Nonvolatile	20.1 %
Binder	10.8 %
Weight per Gallon (lbs/gal)	8.4
VOC (lbs/gal)	2.4
Gloss, 20 deg	90
DOI	84
Adhesion	good
Film Thickness (mil)	0.34
Humidity	good
Water Spray	excell.
Metal Control	excell.

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EXAMPLE 21- Preparation of a Water Borne Silver Metallic Base Coat

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Component A	16.0g
Aluminum Base	
Component B	54.0g
Unpigmented Base	
Component C	85.0g
Reducer Base	

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The above formula was used to prepare a silver metallic base coat. The order of addition is that listed. After each introduction of a component, the entire solution was stirred. The base coat was sprayed on panels primed with an epoxy primer. The base coat was sprayed on the substrate using conventional spray equipment until hiding was achieved. The base coat was allowed to flash for 30 minutes and then clear coated with a transparent coating based on isocyanate technology. The final properties of the base coat after coating with a conventional polyurethane clear coat are listed below, including appearance ratings.

Properties of the Silver Metallic Base Coat and Base Coat, Clear Coat System

Property	Value
Water	73.1 %
Volatile Organic	7.7 %
Pigment	2.7 %
Nonvolatile	19.2 %
Binder	11.2 %
Weight/Gallon (lbs/gal)	8.4
VOC (lbs/gal)	2.6
Gloss, 20	89
DOI	90
Adhesion	good
Film Thickness (mil)	0.53
Humidity	good
Water Spray	excell.
Metal Control	excell.

#### Example 22: Base Coat Preparation

The hydroxyl-functional acrylic polymer is prepared as follows.

Approximately 695 grams of Toluene is combined with 175 grams of Urethane Grade Methyl Ethyl Ketone in a five-liter distillation flask, fitted with a water-cooled condenser, under a Nitrogen blanket. The mixture is brought to reflux, and adjusted to reflux at 100° C by adding Methyl Ethyl Ketone, the Nitrogen flow reduced to a minimum, and a 2215 grams of mixture of  $\beta$ -ethylenically unsaturated monomers is pumped into the flask over two hours. The monomer mixture consists of Methyl Methacrylate - 8.1%, Styrene - 21.2%, n-Butyl Methacrylate - 54.2%, Acrylic Acid - 2.7%, 2-Ethylhexyl Acrylate - 2.8%, and alpha-Methyl Styrene - 2.0%, and Hydroxyethyl Methacrylate - 9.1%. While the monomers are being pumped into the reaction flask, a mixture of 55 grams of free-radical initiator ( ,'-azobis(isobutyronitrile)) and 250 grams solvent (Methyl Ethyl Ketone) is simultaneously fed into the flask over two hours.

After adding all the monomers, solvent, and initiator (and flushing the pumps and feed lines with propylene glycol monopropyl ether), the reaction is held at reflux. After sixty minutes at reflux, a mixture of 11 grams of initiator and 120 grams of solvent (Methyl Ethyl Ketone) is pumped into the reaction flask over seventy-five minutes. After adding the initiator and solvent mixture, the reaction is held at reflux. After one to two hours at reflux, the reaction vessel is cooled to just below reflux, and fitted with a Barrett trap below the condenser. Solvent is distilled off, until 130 grams of solvent is collected. Then the solution is cooled and poured into a lined metal can. The acid number of this polymer is between 19 and 21 milligrams of KOH per gram of resin solids, and its percent non-volatile is between 66% and 69%.

The diisocyanate / polyether adduct is formed by the reaction of one isocyanate group of a diisocyanate with the hydroxyl group of a Methoxy Polyethylene Glycol of the formula:  $\text{CH}_3\text{O}-(\text{CH}_2\text{CH}_2\text{O})_n\text{H}$ , where  $n = 7 - 45$ . This reaction is carried out in a dry air atmosphere, as follows.

In a 1-liter distillation flask, fitted with a water-cooled condenser and a dry air-sparge tube, under a dry air atmosphere, 231.7 grams of isophorone diisocyanate is combined with 35.9 grams of Urethane Grade Methyl Ethyl Ketone, and 374.5 grams of Methoxy Polyethylene Glycol of the formula:  $\text{CH}_3\text{O}-(\text{CH}_2\text{CH}_2\text{O})_n\text{H}$  (molecular weight = 350). The mixture is then stirred for 15 minutes at ambient temperature, and 0.2 grams of dibutyltin diacetate is added. The mixture is continuously stirred, and heated as necessary, to maintain a temperature of 60° C for two hours. The percent isocyanate of the adduct solution should be 6.7% - 6.9%, or 7.2% based on solids.

To make the final urethane / acrylic graft copolymer, 610.9 grams of the hydroxyl-functional acrylic

polymer is combined with 94.6 grams of the diisocyanate / polyether adduct, and 60 grams of Toluene in a five-liter distillation flask, fitted with a water-cooled condenser, under a dry Nitrogen atmosphere. The mixture is heated to 75° C, and held at that temperature for two and one-half hours. The temperature is then increased until the solution reaches reflux (~ 104° C), and held at reflux for thirty minutes. The solution is then cooled to 75° C, and vacuum stripped (85 mbars pressure) to remove approximately 100 grams of solvent. The vacuum is broken with Nitrogen, and 220 grams of ethylene glycol monobutyl ether is added to the flask. After stirring the mixture for ten minutes, vacuum stripping is resumed. Stripping continues at 75° C / 85 mbars until a total of 210 to 212 grams (including the first 100 grams) of solvent has been removed. The solution is then cooled, and filtered into a lined can. The acid number is between 15 and 17 milligrams of KOH per gram of resin solids, and the percent non-volatile is between 66% and 69%.

#### Preparation of Component A - Aluminum Base

A commercial solvent borne aluminum paste, 60.0g, was added to 60.0g of butoxyethanol. The mixture was stirred until a homogeneous dispersion was produced. To this dispersion, 30.0g, of the resin described above was added to the dispersion and the entire mixture stirred until a smooth, uniform slurry was produced. This aluminum base contains 39% volatile organic solvent by weight, a solids level of 61.0% by weight, a binder level of 13% by weight and an aluminum flake level of 26.0% by weight.

#### Preparation of Component B - Pigmented Base

The following pigment slurry was prepared, 12.2g of NeoCryl A-6037, 14.8g of NeoCryl A-640, 0.3g of 30% aqueous ammonia, 3.2g of Perrindo Maroon R-6424 manufactured by Mobay Corporation, 0.1g of a defoamer and 0.05g of Surfynol 104BC manufactured by Air Products. The above mixture were mixed together and predispersed using a Cowles blade. The mixture was then ground on a sand mill until the desired particle size was achieved. A let down solution containing 29.5g of NeoCryl A-622, 2.6g of butoxyethanol, 1.3g of deionized water and 0.03g of 1% aqueous solution of FC-120 was added to the grind portion under agitation. An unpigmented base can be prepared using the procedure above with the elimination of the Perrindo Maroon R-6424 pigment. A wax dispersion was prepared by grinding 50.0g of a wax, 50.0g of unpigmented resin solution, 50.0 g of deionized water and 50.0g of butoxyethanol. This paste was ground to the desired particle size using conventional techniques. To the pigmented and unpigmented bases described above, 6.0g of this wax dispersion was added with stirring. The resulting pigmented base had a pH of 8.6, a solids level of 37.8% by weight, a volatile organic level of 10.6% by weight, a binder level of 33.3% by weight, a pigment level of 5.0% by weight and a water level of 51.6% by weight. The unpigmented base has a solid level of 34.7% by weight, volatile organic level of 9.4% by weight, a water level of 55.9% by weight and a pH of 8.7.

#### Preparation of Component C - Reducer Base

To 1953.2g of deionized water, 31.3g of Laponite RD manufactured by Laporte Industries was added. The resulting solution was then agitated for one hour until the Laponite RD was completely dispersed. To this solution was added 8.0g of Troysol LAC manufactured by Troy Chemical Company and the entire solution stirred for 15 minutes. The resulting solution was then diluted with 1953.2g of deionized water. To this Laponite RD dispersion was added 462.9g of NeoRez R-9637 manufactured by ICI Resins. The resulting mixture had a Laponite RD level of 0.71% by weight, Troysol LAC level of 0.18% by weight, water dispersible polyurethane level of 4.2% by weight and a volatile organic level of 1.1% by weight.

#### EXAMPLE 23 - PREPARATION OF A WATER BORNE RED METALLIC BASE COAT

5.0

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Component	Parts by Weight
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Component A	16.0
Aluminum Base	
Component B	44.0
Pigmented Base	
Component C	95.0
Reducer Base	

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The above formula was used to prepare a red metallic base coat. The order of addition is that listed. After each introduction of a component, the entire solution was stirred. The base coat was sprayed on panels primed with an epoxy primer. The base coat was sprayed on the substrate using conventional spray equipment until hiding was achieved. The base coat was allowed to flash for 30 minutes and then clear coated with a transparent coating based on isocyanate technology. The final properties of the base coat after coating with a conventual polyurethane clear coat are listed below, including appearance ratings.

Properties of the Red Metallic Base Coat and Base Coat, Clear Coat System

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Property	Value
Water	72.5 %
Volatile Organic	7.4 %
Pigment	2.4 %
Nonvolatile	20.1 %
Binder	10.8 %
Weight per Gallon (lbs/gal)	8.4
VOC (lbs/gal)	2.2
Gloss, 20 deg	92
DOI	87
Adhesion	good
Film Thickness (mil)	0.29
Humidity	good
Water Spray	excell.
Metal Control	excell.

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EXAMPLE 24 - Preparation of a Water Borne Silver Metallic Base Coat

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5	Component A	16.0g
	Aluminum Base	
	Component B	44.0g
10	Unpigmented Base	
	Component C	95.0g
	Reducer Base	

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15  
The above formula was used to prepare a silver metallic base coat. The order of addition is that listed.  
20 After each introduction of a component, the entire solution was stirred. The base coat was sprayed on panels primed with an epoxy primer. The base coat was sprayed on the substrate using conventional spray equipment until hiding was achieved. The base coat was allowed to flash for 30 minutes and then clear coated with a transparent coating based on isocyanate technology. The final properties of the base coat after coating with a conventional polyurethane clear coat are listed below, including appearance ratings.

25 Properties of the Silver Metallic Base Coat and Base Coat, Clear Coat System

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30	Property	Value
	Water	73.1 %
	Volatile Organic	7.7 %
	Pigment	2.7 %
	Nonvolatile	19.2 %
40	Binder	11.2 %
	Weight/Gallon (lbs/gal)	8.45
	VOC (lbs/gal)	2.4
	Gloss, 20	93
	DOI	86
45	Adhesion	good
	Film Thickness (mil)	0.37
	Humidity	good
	Water Spray	excell.
50	Metal Control	excell.

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55 Claims

1. A metallic automotive paint base coat kit composition consisting of:



- a) an aluminum base containing a water reducible acrylic resin having nonionic and acid functionality,
- b) pigmented base containing an acrylic latex resin and,
- c) reducer base containing a rheology control agent.

2. The base coat kit composition of claim 1 wherein the aluminum base contains a water reducible solvent borne acrylic resin having nonionic and acid functionality resulting from the solution polymerization of ethylenically unsaturated monomers wherein at least two ethylenically unsaturated monomers are acid and poly(ethylene oxide) functional and optionally ethylenically unsaturated monomer containing hydroxyl groups to give a polymer with a glass transition temperature from about -20 to 50 °C (using the Fox method of Tg calculation), relative weight average molecular weight of 1,00 to 50,000, an acid number of 5 to 50, a hydroxyl number of 0 - 50, and a final poly(ethylene oxide) level of 1 - 50 % by weight.

3. The base coat kit composition of claim 1 wherein the aluminum base contains a water reducible acrylic resin having nonionic, urethane and acid functionality comprising:

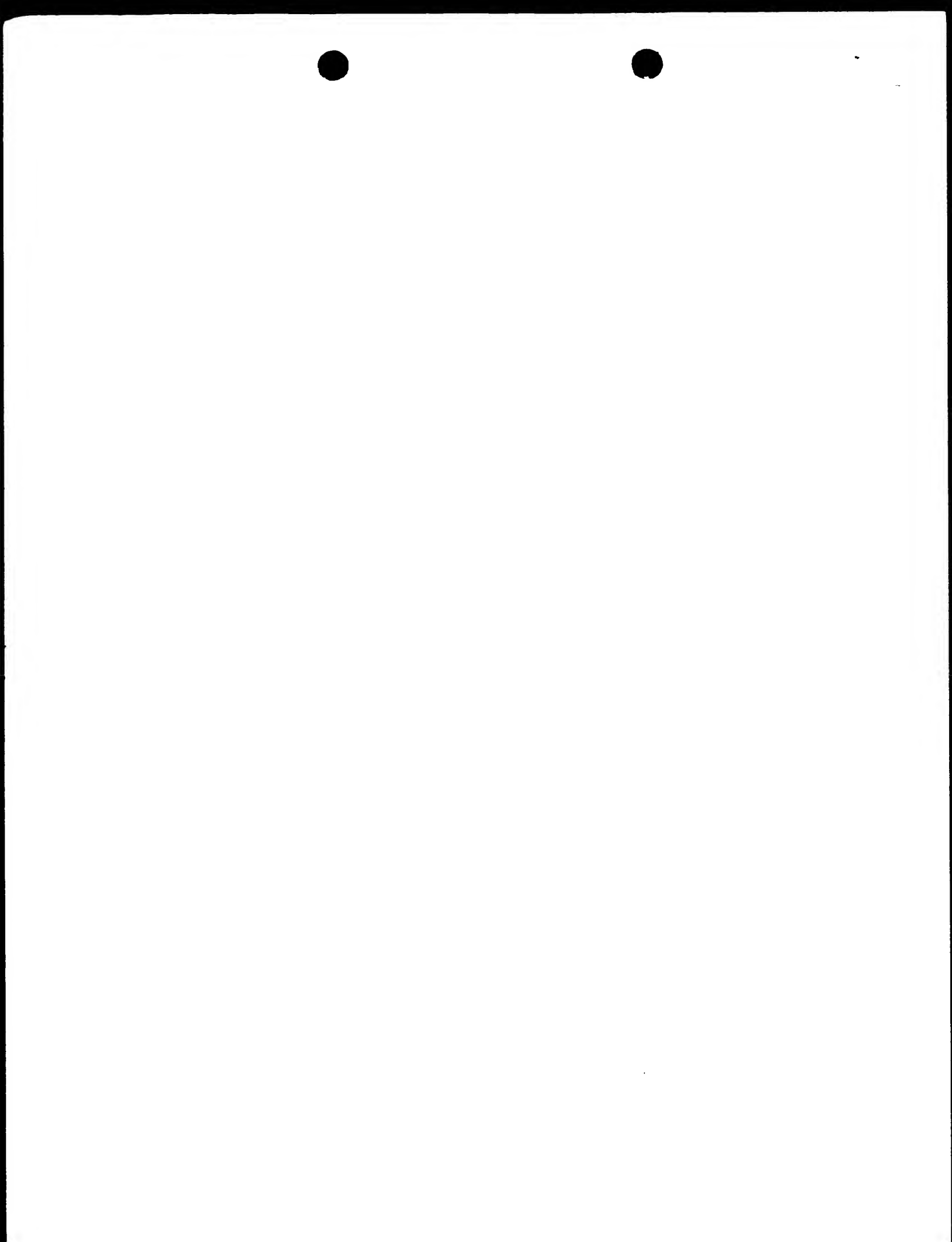
- a) at least one monomer having acid functionality,
- b) at least one monomer having urethane and nonionic functionality prepared by the reaction of:
  - 1. a capped polyether having a single active hydrogen functionality with a polyisocyanate compound to produce a partially capped isocyanate intermediate which is further reacted with a ethylenically unsaturated compound having at least one active hydrogen group to produce an ethylenically unsaturated compound having urethane and nonionic functionality, or
  - 2. a capped polyether having a single active hydrogen functionality with a polyisocyanate compound having ethylenically unsaturated functionality to produce an ethylenically unsaturated compound having urethane and nonionic functionality,
- c) ethylenically unsaturated compound
- d) and optionally ethylenically unsaturated compounds having active hydrogen groups.

4. The base coat kit composition of claim 1 wherein the aluminum base contains a water reducible acrylic resin having nonionic, urethane and acid functionality prepared by the reaction of an acrylic resin having active hydrogen functionality with an isocyanate adduct having urethane and nonionic functionality.

5. A water reducible acrylic resin having nonionic, urethane and acid functionality suitable for the use in metallic automotive paint base coat compositions comprising:

- a) at least one monomer having acid functionality,
- b) at least one monomer having urethane and nonionic functionality prepared by the reaction of:
  - 1. a capped polyether having a single active hydrogen functionality with a polyisocyanate compound to produce a partially capped isocyanate intermediate which is further reacted with an ethylenically unsaturated compound having at least one active hydrogen group to produce an ethylenically unsaturated compound having urethane and nonionic functionality, or
  - 2. a capped polyether having a single active hydrogen functionality with a polyisocyanate compound having ethylenically unsaturated functionality to produce an ethylenically unsaturated compound having urethane and nonionic functionality,
- c) ethylenically unsaturated compound,
- d) and optionally ethylenically unsaturated compounds having active hydrogen groups.

6. A water reducible acrylic resin having nonionic, urethane and acid functionality suitable for the use in metallic automotive paint base coat compositions prepared by the reaction of an acrylic resin having active hydrogen functionality with an isocyanate adduct having urethane and nonionic functionality.



(19)



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(4) **Metallic water borne base coat composition.**

(5) There is disclosed a metallic and nonmetallic water borne base coat compositions exhibiting excellent appearance. A method for long term stability and the introduction of aluminum flake pigments into a water borne coating is also disclosed. The method is based on storing the aluminum flake in a solvent environment with introduction of the aluminum flake in to the aqueous environment just before applying the base coat coating. A water reducible acrylic resin with nonionic and acid functionality has been disclosed that allows for direct introduction of aluminum flake in to an aqueous environment. The coating vehicle is based

on acrylic latex vehicle which exhibits excellent drying and film properties. The coating compositions disclosed are in compliance with current and proposed volatile organic compound regulations.

EP 0 471 972 A3



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## EUROPEAN SEARCH REPORT

Application Number

EP 91 11 1567

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.5)
P, X	EP-A-0 424 705 (BASF) * page 3, line 40 - page 7, line 5; claims 1-15; example W *	1	C08G18/81 C08G18/62 C09D133/06 C09D5/38 B05D7/16 C09D175/08
D, X	EP-A-0 287 144 (AKZO) * page 2, line 22 - page 4, line 36; claims 1-11 *	1	
A	EP-A-0 326 984 (VIANOVA) * page 2, line 36 - page 3, line 2; claims 1-10; example 1 *	1	
A	EP-A-0 354 261 (BASF) * page 3, line 5 - page 10, line 49; claims 1-36; example 7 *	1-3	
A	US-A-4 719 132 (PORTER) * column 2, line 35 - column 7, line 58; claims 1-12; examples H, J, K *	1	
A	EP-A-0 365 098 (AKZO) * page 3, line 29 - page 6, line 26; claims 1-23 *	1-6	TECHNICAL FIELDS SEARCHED (Int. Cl.5)
A	GB-A-2 021 599 (UCC) * claims 1-14 *	1-6	C08G C09D B05D
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 11 MARCH 1992	Examiner BOURGONJE A. F.
<b>CATEGORY OF CITED DOCUMENTS</b> X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons Δ : member of the same patent family, corresponding document			